

1 **Investigating sources of measured forest-atmosphere**
2 **ammonia fluxes using two-layer bi-directional modelling**

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17

18 **Abstract**

19 Understanding and predicting the ammonia (NH₃) exchange between the biosphere and the
20 atmosphere is important due to the environmental consequences of the presence of reactive
21 nitrogen (N_r) in the environment. The dynamics of the natural sources are, however, not well
22 understood, especially not for forest ecosystems due to the complex nature of this soil-
23 vegetation-atmosphere system. Furthermore, the high reactivity of NH₃ makes it technically
24 complex and expensive to measure and understand the forest-atmospheric NH₃ exchange. The
25 aim of this study is to investigate the NH₃ flux partitioning between the ground layer, cuticle
26 and stomata compartments for two temperate deciduous forest ecosystems located in
27 Midwestern, USA (MMSF) and in Denmark (DK-Sor). This study is based on measurements

1 and simulations of the surface energy balance, fluxes of CO₂ and NH₃ during two contrasted
2 periods of the forest ecosystems, a period with full developed canopy (MMSF) and a
3 senescent period for the DK-Sor site, with leaf fall and leaf litter build-up. Both datasets
4 indicate emissions of NH₃ from the forest to the atmosphere. The two-layer NH₃
5 compensation point model SURFATM-NH₃ was used in combination with a coupled
6 photosynthesis-stomatal conductance model to represent seasonal variation in canopy
7 physiological activity for simulating both net ecosystem CO₂ exchange rates ($R^2 = 0.77$ for
8 MMSF and $R^2 = 0.84$ for DK-Sor) and atmospheric NH₃ fluxes ($R^2 = 0.43$ for MMSF and R^2
9 $= 0.60$ for DK-Sor). A scaling of the ground layer NH₃ emission potential (Γ_g) was
10 successfully applied using the plant area index (*PAI*) to represent the build-up of a litter layer
11 in the leaf fall period. For a closed green forest canopy (MMSF), unaffected by agricultural
12 NH₃ sources, NH₃ was emitted with daytime fluxes up to 50 ng NH₃-N m⁻² s⁻¹ and nighttime
13 fluxes up to 30 ng NH₃-N m⁻² s⁻¹. For a senescing forest (DK-Sor), located in an agricultural
14 region, deposition rates of 250 ng NH₃-N m⁻² s⁻¹ were measured prior to leaf fall, and
15 emission rates up to 670 ng NH₃-N m⁻² s⁻¹ were measured following leaf fall. For MMSF,
16 simulated stomatal NH₃ emissions explain the daytime flux observations well, and it is
17 hypothesized that cuticular desorption is responsible for the observed NH₃ emissions at night.
18 During leaf fall in DK-Sor, ground fluxes dominate the NH₃ flux with a mean emission rate of
19 150 ng NH₃-N m⁻² s⁻¹. This study shows that forests potentially comprise a natural source of
20 NH₃ to the atmosphere, and that it is crucial to take into account the bi-directional exchange
21 processes related to both the stomatal, cuticular and ground layer pathways in order to
22 realistically simulate forest–atmosphere fluxes of NH₃.

23

24 Keywords: ammonia; biosphere atmosphere exchanges; compensation point; deciduous
25 forest; measurements; modelling

26

27 **1 Introduction**

28 Predicting the surface exchange of atmospheric ammonia (NH₃) is important in order to
29 assess the environmental consequences of the presence of reactive nitrogen (N_r) in the
30 environment (Sutton et al. 2011). However, prediction of the NH₃ exchange between the
31 biosphere and the atmosphere with process-based models is challenging due to the complex
32 nature of the soil-vegetation-atmosphere system (e.g., Sutton et al. 2013). These exchange

1 processes are controlled by a number of feedback mechanisms depending on climatic,
2 biological, chemical and physical conditions (Flechard et al. 2013).

3 Atmospheric chemistry and transport models (CTMs) are recognized tools for studying the
4 fate of nitrogen (N) in the coupled biosphere – atmosphere system (Bash et al. 2013; Pinder et
5 al. 2008; Rao et al. 2011; Tuccella et al. 2012; Wichink Kruit et al. 2012). In the past decade,
6 these models have been improved substantially to represent the governing processes that
7 determine atmospheric NH₃ fluxes (e.g., Hertel et al. 2012; Hamaoui-Laguel et al. 2014;
8 Hendriks et al. 2013). This includes the development of dynamic NH₃ emissions models
9 (Paulot et al. 2014; Skjøth et al. 2011), detailed NH₃ emission inventories (Paulot et al. 2014;
10 Pouliot et al. 2012; Kuenen et al. 2014; Velthof et al. 2012) and the parameterization of
11 processes for simulating canopy NH₃ compensation points (Bash et al. 2013; Wichink Kruit et
12 al. 2012), i.e., the atmospheric NH₃ concentrations at which the net atmospheric NH₃ flux is 0
13 ng m⁻² s⁻¹.

14 Emissions of atmospheric NH₃ are mainly related to agriculture (Reis et al. 2009), generally
15 as a result of volatilizations from animal husbandry, the storages and spreading of manure and
16 mineral fertilizer (Skjøth and Geels 2013) that were found to be the dominant drivers of the
17 spatial and temporal atmospheric NH₃ concentrations (Hertel et al. 2012; Sutton et al. 2013).
18 However, NH₃ emissions also occur from natural sources such as from wild animals (e.g.,
19 Riddick et al. 2014; Theobald et al. 2006), forest fires (e.g., Andreae and Merlet 2001; Van
20 Damme et al. 2014b), sea surfaces (e.g., Sørensen et al. 2003), terrestrial ecosystems (e.g.,
21 Andersen et al. 1999; Hansen et al. 2013; Sutton et al. 1997), and from the chemical
22 partitioning of N compounds between the gas and aerosol phases (Pryor et al. 2001), but the
23 dynamics of these natural sources are not well understood, especially not for unmanaged
24 ecosystems (Erisman and Wyers 1993; Hansen et al. 2013; Sutton et al. 1997; Wang et al.
25 2011).

26 Atmospheric NH₃ exchange with the biosphere is bi-directional and it follows several
27 pathways; the soil, the leaves cuticles, and the stomata (e.g., Nemitz et al. 2001). Usually, it
28 has been assumed that NH₃ deposition occurs onto leaf surfaces and natural NH₃ emissions
29 occurs through the stomata depending on a stomatal NH₃ compensation point (Farquhar et al.
30 1980a). However, the fluxes can be bi-directional for all the compartments and depend on the
31 concentration difference between the atmosphere and the compartment. Each compartment
32 has a varying (unitless) NH₃ emission potential (*I*) which is defined as the ratio of ammonium

1 (NH₄⁺) to hydrogen (H⁺) ions in the water (Schjoerring et al. 1998). Usually the ground is the
2 main source of NH₃, especially in agricultural ecosystems which receive large amount of
3 nitrogen (Nemitz et al. 2000a; Sutton et al. 2009; Personne et al. 2015; Ferrara et al. 2014).
4 These emissions may be due to direct emissions due to agricultural operations such as
5 application of slurry (e.g., Ferrara et al. 2016) or may also be due to the microbiological
6 breakdown of leaf litter (Nemitz et al. 2000a; David et al. 2009). Breakdown of litter also
7 happens in non-agricultural systems such as forests where it was found to contribute to
8 ecosystem fluxes of biogenic volatile organic compounds (BVOCs) (Greenberg et al. 2012)
9 contributing to feed-back mechanisms within the Earth system (e.g., Carslaw et al. 2010).
10 Forest NH₃ emissions have been observed in late summer/autumn periods that may be related
11 to litter decomposition and soil evaporation (Hansen et al. 2013; Hansen et al. 2015)
12 indicating that such sources could be relevant to include in CTM models.

13 In bi-directional NH₃ exchange models, the unitless NH₃ emission potentials of the ground
14 layer (Γ_g) and stomata (Γ_s) are required to simulate the NH₃ compensation points of the
15 ground layer and the stomata, respectively (Wichink Kruit et al. 2010). Typically, these
16 models use constant Γ -values based on measurements, however, such measurements yet only
17 exist to a very limited extent and are demanding to conduct. Furthermore, there is a
18 substantial need to represent dynamic growing seasons in existing CTMs (Simpson et al.
19 2012) in order to represent realistic seasonal vegetation and ground layer emissions of
20 nitrogen oxide (NO), NH₃, and BVOCs. During the growing season, physiological and
21 biogeochemical processes cause seasonal variations in photosynthesis, stomatal conductance,
22 leaf development as well as N mobilization and translocation (Wang et al. 2013). These
23 processes are affecting the stomatal emission potential (Wang et al. 2011) and stomatal
24 conductance being strongly correlated with both NH₃ emission and deposition fluxes of leaves
25 (Gessler et al. 2000). Furthermore, seasonal variation includes the dynamic development of a
26 leaf litter layer and decomposition influencing the ground layer emission potential (Callesen
27 et al. 2013).

28 The aim of this study was to investigate the contribution of the leaves and forest floor to the
29 net NH₃ exchange at different development stages of the forest including fully developed and
30 senescing periods. By using the two-layer bi-directional exchange model SURFATM-NH₃
31 (Personne et al. 2009) as a comparison and interpretation tool, the simulated fluxes are
32 evaluated for two temperate deciduous forest reported by Hansen et al. (2013 and 2015). First,

1 the partitioning and the temporal pattern of the net flux of NH₃ measured above the two
2 temperate deciduous forests are presented and then, the sources of NH₃ are interpreted and
3 discussed in relation to the phenological state of the forest canopies.

4 5 **2 Methods**

6 **2.1 Experimental data and sites**

7 The atmospheric NH₃ concentration and fluxes of NH₃, carbon dioxide (CO₂) and heat were
8 measured at two temperate deciduous forest sites, a beech forest study site in Denmark (DK-
9 Sor) for 25 days during late fall in 2010 (Hansen et al. 2013), and the Morgan-Monroe State
10 Forest (MMSF) site in the central Midwestern USA for 5 days during late summer in 2013
11 (Hansen et al. 2015) (Table 1). The atmospheric NH₃ measurements were conducted with
12 half-hourly temporal resolution using the Relaxed Eddy Accumulation (REA) method
13 (Businger and Oncley 1990) in combination with Wet Effluent Diffusion Denuders (WEDD)
14 (Sørensen et al. 1994).

15 **2.1.1 The DK-Sor site**

16 The DK-Sor forest is located in the central part of Zealand (55°29'N, 11°38'E). The forest
17 consists predominantly of beech trees (*Fagus sylvatica* L.) with an average canopy height of
18 26 m and covers 2.5 km². The mean summer peak plant area index (*PAI*) in the period 2000 to
19 2011 was 4.6 m² m⁻² with maximum *PAI* just above 5 m² m⁻² (Pilegaard et al. 2011). The
20 surrounding landscape is dominated by agricultural land use. During the 25 day measurement
21 campaign (21 October to 15 November 2010), the measured forest canopy *PAI* (LAI-2000,
22 Li-Cor, USA) decreased from 3.7 m² m⁻² to 1.1 m² m⁻² and the mean temperature was 6.7°C.
23 Leaf fall ended on 8 November (Hansen et al. 2013) where *PAI* equalizing 1.1 m² m⁻²
24 representing trunks and branches only.

25 **2.1.2 The MMSF site**

26 Morgan-Monroe State Forest (MMSF) is located at 39°53'N, 86°25'W in Southern Indiana,
27 USA. MMSF is a secondary successional broadleaf forest dominated by the deciduous tree
28 species tulip poplar (*Liriodendron tulipifera*), white oak (*Quercus alba*), sassafras (*Sassafras*
29 *albidum*), and sugar maple (*Acer saccharum*) and covers 97 km². The canopy height is 28-30

1 m and the summer peak *PAI* during 2013 was $4.6 \text{ m}^2 \text{ m}^{-2}$ (Hansen et al. 2015). Beyond the
2 limits of the forest, the surrounding land cover is dominated by cropland. During the 5-day
3 measurement campaign (5 September to 10 September), *PAI* was $4.5 \text{ m}^2 \text{ m}^{-2}$ and the mean
4 temperature was 24.5°C .

5 2.1.3 CO₂ and energy fluxes

6 Carbon dioxide and energy flux observations and meteorological data used for input to the
7 models or model validation were obtained from the European and American Fluxes Database
8 Clusters; FluxNet (www.europe-fluxdata.eu) and AmeriFlux (<http://ameriflux.ornl.gov>). Eddy
9 covariance data were gap-filled, flux-partitioned, and friction velocity (u^*) corrections were
10 applied according to the standard procedure in FluxNet (Papale et al. 2006; Reichstein et al.
11 2005).

12 The energy flux data for the DK-Sor site needed to be filtered due to a sensitivity of the sonic
13 anemometer that made sonic temperature fluctuation measurements at high wind speeds
14 unreliable. Data points during periods with high wind speeds ($> 5 \text{ m s}^{-1}$) (DOY 294–300 and
15 307–310) were therefore removed.

16 2.2 The SURFATM-NH₃ model

17 The SURFATM-NH₃ model (Personne et al. 2009) is a one-dimensional model that uses a
18 two-layer bi-directional NH₃ exchange scheme including a stomatal and ground layer NH₃
19 compensation point. By coupling a water and energy balance model with the two-layer NH₃
20 resistance scheme, SURFATM-NH₃ simulates the atmospheric NH₃ flux based on measured
21 atmospheric NH₃ concentrations, and meteorological and vegetation input (net radiation, soil
22 and air temperature, relative air humidity, wind speed, *PAI* and rain. The model furthermore,
23 uses predefined NH₃ emission potentials for stomata and the ground layer of the site. The
24 scheme is based on the traditional resistance analogue describing the bi-directional transport
25 of NH₃ governed by a set of resistances controlled by the atmosphere, r_a (s m^{-1}), the quasi-
26 laminar boundary layer, r_b (s m^{-1}), and the canopy, r_c (s m^{-1}) respectively (e.g., Erisman and
27 Wyers 1993). It expands the existing one-layer canopy NH₃ compensation point model
28 (Sutton et al. 1998) with a ground layer compensation point, χ_g (mol l^{-1}), allowing emissions
29 from the ground layer (Nemitz et al. 2001) (see Appendix A). In a similar way to the stomatal
30 NH₃ compensation point, χ_s (mol l^{-1}), χ_g is estimated from the Henry's law and dissociation

1 constants ($K_H = 10^{-1.76}$ (unitless) and $K_d = 10^{-9.25} \text{ mol L}^{-1}$) (Equation 1) and the dimensionless
 2 emission potential of the ground layer (Γ_g) (Equation 2).

$$3 \quad \chi_g = \Gamma_g \times K_d \times K_H \times \exp\left(\frac{\Delta H_H^0 + \Delta H_d^0}{R} \times \left(\frac{1}{298.15} - \frac{1}{T_g}\right)\right) \quad (1)$$

$$4 \quad \Gamma_g = [\text{NH}_4^+] / [\text{H}^+] \quad (2)$$

5 with ΔH_H^0 and ΔH_d^0 being free enthalpies of acid-base dissociation $\text{NH}_4^+/\text{NH}_3$ (kJ mol^{-1}) and for
 6 NH_3 volatilization (kJ mol^{-1}) (Personne et al. 2009), R ($0.00831 \text{ kJ K}^{-1} \text{ mol}^{-1}$) is the perfect
 7 gas constant, and T_g (K) is the temperature of the ground layer. The model simulates the total
 8 net atmospheric NH_3 flux, F_T ($\mu\text{g m}^{-2} \text{ s}^{-1}$) as a sum of each of the forest component fluxes; the
 9 stomatal, F_s ($\mu\text{g m}^{-2} \text{ s}^{-1}$), cuticular, F_w ($\mu\text{g m}^{-2} \text{ s}^{-1}$), and ground, F_g ($\mu\text{g m}^{-2} \text{ s}^{-1}$), flux which are
 10 all related to the NH_3 canopy compensation point, χ_c ($\mu\text{g m}^{-3}$) (see Appendix A).

11 **2.3 Model setup**

12 The SURFATM- NH_3 model runs with a set of initialized state variables, physical parameters
 13 and constants (Table 2). As SURFATM- NH_3 was formerly applied for agricultural sites
 14 (Personne et al. 2015; Loubet et al. 2012), model parameters were adjusted to represent the
 15 two forest sites. When available, field measurements were used to set or calculate parameter
 16 values, or parameters were taken from published scientific work carried out at the sites (see
 17 references in Table 2). Otherwise, theoretical values were used (see references in Table 2), or
 18 parameters were estimated by trial-error method within a range of realistic values found in the
 19 scientific literature (see Table 2).

20 **2.3.1 Modelling the stomatal resistance**

21 A physiologically based leaf photosynthesis-stomatal conductance model approach proposed
 22 by Collatz et al. (1991) was used to simulate stomatal resistance (r_s). Based on dynamic
 23 coupling between a stomatal conductance, g_s (m s^{-1}) model formulated by Ball et al. (1987)
 24 (Equation 4) and mechanistic simulations of photosynthesis (Equation 5), the stomatal
 25 resistance ($r_s = 1/g_s$) was simulated and included in SURFTAM- NH_3

$$26 \quad g_s = m \frac{A_n h_s}{C_s} + b \quad (4)$$

$$27 \quad A_n = \min \{J_E, J_C, J_S\} - R_D \quad (5)$$

1 The relative humidity at leaf surface, h_s (unitless), and the CO_2 partial pressure at leaf surface,
2 C_s (Pa), determines g_s along with leaf scale net carbon assimilation, A_n ($\text{mol m}^{-2} \text{s}^{-1}$), and two
3 fixed constants ($m = 7$ and $b = 0.01 \text{ mol m}^{-2} \text{ s}^{-1}$) representing the slope and intercept. A_n is
4 determined by the minimum of three potential capacities and the leaf dark respiration, R_D
5 ($\text{mol m}^{-2} \text{ s}^{-1}$) following Equation 5 (Collatz et al. 1991; Farquhar et al. 1980b). J_E ($\text{mol m}^{-2} \text{ s}^{-1}$)
6 is the light-limited assimilation rate, J_C ($\text{mol m}^{-2} \text{ s}^{-1}$) is the rubisco-limited assimilation rate,
7 and J_S ($\text{mol m}^{-2} \text{ s}^{-1}$) is the assimilation rate due to the limitation of the export of assimilates
8 inside the leaf. Measured PAI is used as model input to upscale leaf simulations to canopy
9 scale (Sellers et al. 1992). Details of the coupled photosynthesis-stomatal conductance model
10 and the soil/ecosystem respiration parameterization are described in Appendix B. The results
11 from using the g_s model were evaluated using measured eddy covariance CO_2 fluxes
12 (Pilegaard et al. 2011; Schmid et al. 2000) to verify NEE simulations. The simulated r_s
13 estimates were then utilized for modelling the atmospheric NH_3 exchange rate using
14 SURFATM- NH_3 . The simulations were performed for the full years 2010 (DK-Sor) and 2013
15 (MMSF) to examine the seasonal performance of the model.

16 2.3.2 Emission potential of the ground layer (Γ_g) and the stomata (Γ_s)

17 The NH_3 emission potentials of the ground layer and stomata, Γ_g and Γ_s , were not measured at
18 the two sites during the measurement campaigns. Therefore, measurements of Γ_s from late fall
19 period in 2008 and 2009 from the DK-Sor site, reported by Wang et al. (2011 and 2013), were
20 used to set $\Gamma_s = 200$ for DK-Sor representative of senescing leaves, and a constant value for
21 MMSF of 400 was used to represent Γ_s of a green forest canopy with PAI close to its
22 maximum value. In this paper, we suggest a scaling of Γ_g in the leaf fall period using PAI to
23 represent N enrichment of the ground layer (soil + litter) due to litter fall:

$$24 \quad \Gamma_g = \Gamma_{g,\min} + [(\Gamma_{g,\max} - \Gamma_{g,\min}) \times \Delta LPAI] \quad (6)$$

25 where $\Delta LPAI = 1 - \left(\frac{PAI - PAI_{\min}}{PAI_{\max} - PAI_{\min}} \right)$ represents the change in the litter layer derived from the
26 measured PAI using the LAI-2000 sensor (Figure 1). Considering the lag time from the
27 beginning of leaf fall until decomposition is efficient, the scaling is applied for the period with
28 PAI decreasing from 3.5 (hence $PAI_{\max} = 3.5 \text{ m}^2 \text{ m}^{-2}$) until it reaches its minimum value
29 ($PAI_{\min} = 1.1 \text{ m}^2 \text{ m}^{-2}$). Predefined minimum and maximum values of Γ_g are used. We set $\Gamma_{g,\min}$
30 $= 300$ based on litter measurements from Wang et al. (2011 and 2013), and $\Gamma_{g,\max} = 18000$ is

1 estimated by trial and error method to represent the higher ground layer N emission potential
2 following the leaf fall period (Figure 1).

3

4 **3 Results**

5 **3.1 Model testing for energy and CO₂ fluxes**

6 Before SURFATM-NH₃ was applied to simulate the atmospheric NH₃ fluxes above the two
7 forests, the model's physical representation of the ecosystem dynamics was evaluated by 1)
8 verifying the physiological representation of the canopy by comparing measured and
9 modelled *NEE*, and 2) comparing simulated and measured energy fluxes.

10 SURFATM-NH₃ was run for all days with available NH₃ flux data for this study, i.e., 5 days
11 in the late summer 2013 (DOY 248–253) for the MMSF forest site, and 25 days during the
12 leaf fall period 2010 (DOY 294–319) for the DK-Sor site (Figure 2).

13 **3.1.1 Net ecosystem exchange (*NEE*)**

14 Model simulations of *NEE* are strongly correlated with measured CO₂ flux data (Figure 3) for
15 both MMSF ($R^2 = 0.77$) and DK-Sor ($R^2 = 0.84$), and high concordance correlation
16 coefficients (*CCC*) further signify good agreement between data and simulations of the two
17 sites ($CCC = 0.72$ for MMSF and $CCC = 0.83$ for DK-Sor). For DK-Sor, the stomatal activity
18 was less towards the end of the observation period due to leaf senescence and leaf fall. Hence
19 the modelled atmospheric fluxes were less sensitive to leaf-scale r_s variability in that period.
20 The close agreement of the simulated CO₂ fluxes to the measured CO₂ fluxes ensures the
21 consistent integration of the stomatal resistance r_s in SURFTAM-NH₃ model.

22 **3.1.2 Energy fluxes**

23 During the measurement period of MMSF, the forest *PAI* was 4.5 m² m⁻², the mean
24 temperature was 24.3 °C, and it rained 12.8 mm (Table 1). The rain fell within a 3-4 hour
25 period during the night on DOY 251 (Figure 2). Over the five days, the energy fluxes showed
26 a typical pattern for vegetated ecosystems of peak fluxes during daytime with sensible heat
27 fluxes (*H*) of up to 200 W m⁻² and latent heat fluxes (*LE*) of up to 400 W m⁻². Ignoring in-
28 canopy heat storage and metabolic terms, the average instantaneous energy balance closure
29 fraction ($(H+LE)/(R_n-G)$) was 0.50 (Figure 4a), however accounting for the storage terms is

1 important for the energy balance closure (Stoy et al. 2013). The model simulates the diel
2 patterns and ranges of the energy fluxes in strong agreement with observations, i.e., $R^2 = 0.78$
3 and $CCC = 0.69$ for H , and $R^2 = 0.87$ and $CCC = 0.78$ for LE (Table 3).

4 The measurement period of DK-Sor was characterized by decreasing temperatures, leaf
5 senescence and leaf fall. The canopy PAI decreased from $3.7 \text{ m}^2 \text{ m}^{-2}$ to $1.1 \text{ m}^2 \text{ m}^{-2}$ between
6 DOY 294 and 312, and the mean temperature throughout the period was $6.7 \pm 2.6^\circ\text{C}$ (Table
7 1). LE was continuously lower than 200 W m^{-2} , and H reached a daytime maximum of 100 W
8 m^{-2} only twice for the 25 days of observed period. The simulated reference evaporation (Allen
9 et al. 1998) confirmed low atmospheric evaporative demand (between -50 and 100 W m^{-2})
10 during the rainy and overcast measurement period (Figure 2). Even though the energy balance
11 closure (Figure 4b) and the statistical synthesis for the comparison between simulations and
12 measurements for the DK-Sor site (Table 3) are very weak during this overcast and rainy
13 period (i.e., $R^2 = 0.17$ and $R^2 = 0.07$ for H and LE , respectively), the typical diel pattern of the
14 fluxes (H and LE) is clearly recognized (Figure 2).

15 **3.2 Ammonia fluxes**

16 At MMSF, the fluxes were positive during both day and night, indicating a release of NH_3
17 from the forest ecosystem to the atmosphere. The measured NH_3 fluxes showed a clear day-
18 time pattern with maximum emissions during midday of up to $51.6 \text{ ng NH}_3\text{-N m}^{-2} \text{ s}^{-1}$ (Figure
19 5a), and the model represented the same day-time pattern with peak emissions during midday.
20 The simulated range of daytime NH_3 emissions is also in good agreement with measurements
21 during most of the period (between 36 and $46 \text{ ng NH}_3 \text{ m}^{-2} \text{ s}^{-1}$), however, the NH_3 fluxes are
22 overestimated during midday on the last two days. During nighttime, the model simulated
23 zero or negative net NH_3 exchange, while emissions of up to $30 \pm 70 \text{ ng NH}_3\text{-N m}^{-2} \text{ s}^{-1}$ were
24 measured.

25 The measured NH_3 fluxes for DK-Sor show deposition fluxes of $-250 \pm 300 \text{ ng NH}_3\text{-N m}^{-2}$
26 s^{-1} in the beginning of the period that gradually change to emission fluxes of up to 670 ± 280
27 $\text{ng NH}_3\text{-N m}^{-2} \text{ s}^{-1}$ towards the end of the measurement period (Figure 5b). This change
28 occurred due to leaf senescence and leaf fall causing a smaller canopy surface area for NH_3
29 depositions and possibly NH_3 emissions related to N translocation and soil emissions (Hansen
30 et al. 2013). Contrary to NH_3 flux measurements at MMSF, no clear diurnal variation was
31 observed in NH_3 fluxes during the leaf-fall period in DK-Sor. Fluxes turned from negative

1 (deposition) to positive (emission) on DOY 303 at which time *PAI* had decreased from 4 m²
2 m⁻² to less than 3 m² m⁻². The mean emission rate was 150 ± 138 ng NH₃-N m⁻² s⁻¹ during the
3 rest of the measurement period (until DOY 319). The model simulated well the measured
4 NH₃ emissions following leaf fall, however, during DOY 314–316, a NH₃ emission event
5 with fluxes up to 500 ± 131 ng NH₃-N m⁻² s⁻¹ was measured which was not captured by the
6 model.

7 **3.3 Ammonia flux contributions**

8 The SURFATM-NH₃ model was used to analyze the contribution of the individual sources to
9 the total flux. It was found that for MMSF (Figure 6a), the stomatal exchange was the main
10 contributor (up to 50 ng NH₃-N m⁻² s⁻¹) to the simulated forest NH₃ emissions during
11 daytime. The strong stomatal control of NH₃ emissions is in turn controlled by environmental
12 factors with a strong diel signal (radiation, temperature, humidity, CO₂). The modelled
13 deposition to the leaf cuticles was small (up to 4.5 ng NH₃-N m⁻² s⁻¹) and predominant during
14 the night when the other components were less active, and relative air humidity was high. The
15 modelled ground layer only contributed with small emissions during day time (up to 1.5 ng
16 NH₃-N m⁻² s⁻¹). The observed nighttime emissions were not simulated by the model.

17 For DK-Sor, during leaf fall (Figure 6b), the diurnal pattern differed substantially from that of
18 the green canopy of MMSF. Here, the ground layer, or more specifically the fresh
19 decomposing litter layer, was by far the largest contributor (up to 150 ng NH₃-N m⁻² s⁻¹) to
20 the total simulated NH₃ emissions from the forest to the atmosphere. Depositions (up to 30 ng
21 NH₃-N m⁻² s⁻¹) to the cuticular surfaces were simulated for DK-Sor whereas stomata were
22 inactive during most of the measurement period due to advanced leaf senescence, and hence
23 did not contribute significantly to the regulation of the NH₃ flux. The NH₃ fluxes thus showed
24 a less pronounced diel pattern (Figure 6b) with slightly higher emissions during daytime
25 (average of 91 ng NH₃-N m⁻² s⁻¹) as compared to nighttime (average of 70 ng NH₃-N m⁻²
26 s⁻¹).

27 **3.4 Model sensitivity to the emission potentials**

28 The sensitivity of the simulated mean diel NH₃ flux to the emission potentials for leaves (I_s)
29 and the ground layer (I_g), respectively, was examined for the different phenological stages
30 represented by the two studied forests. For this purpose, a range of 0-1000 was chosen for I_s

1 and 0-30000 for Γ_g as inputs for SURFATM-NH₃ modelling. The modelled NH₃ fluxes of the
2 green forest canopy were sensitive to Γ_s (Figure 7a), but this was not the case for the
3 senescing forest canopy (Figure 7b) with *PAI* decreasing from 3.7 m² m⁻² to 1 m² m⁻² (Figure
4 1). During and after leaf fall, the modelled NH₃ fluxes of DK-Sor were very sensitive to Γ_g ,
5 while the sensitivity of NH₃ fluxes to the large range of Γ_g input values is less for MMSF
6 (Figure 7c and 7d). Due to the use of *PAI* for scaling Γ_g in this study (Equation 6), Γ_g will
7 however remain low (close to $\Gamma_{g,\min}$) for a green closed canopy such as MMSF, and this
8 causes also the simulated soil NH₃ fluxes to remain low (Figure 6a) irrespective of the
9 parameter value set for $\Gamma_{g,\max}$. In contrast, the simulated NH₃ fluxes of senescent forests (e.g.,
10 DK-Sor) will remain very sensitive to the chosen parameter value for $\Gamma_{g,\max}$, and the soil NH₃
11 emission contributes significantly to the canopy NH₃ fluxes in this case (Figure 6b).

12

13 **4 Discussion**

14 This study aimed to analyze contributions of measured NH₃ fluxes from individual forest
15 compartments (ground layer, cuticle and stomata) and to quantify these individual
16 contributions to the net forest - atmosphere NH₃ flux for two deciduous forests showing
17 distinct diurnal (MMSF) and non-diurnal (DK-Sor) NH₃ flux patterns indicative of forest NH₃
18 emissions. The distinct diurnal and non-diurnal flux patterns may be related to dominant
19 processes influencing forest NH₃ emissions in different phenological phases and in different
20 landscape settings. In particular, MMSF is located in a remote region while DK-Sor is located
21 in an agricultural region characterized by large atmospheric NH₃ depositions in the growing
22 season (Hansen et al. 2013). Thus, only MMSF (not DK-Sor) show NH₃ emissions in the
23 green (mid-season) period, and only DK-Sor (not MMSF) show NH₃ emissions in the leaf-fall
24 period (see Hansen et al. 2013; 2015). In order to analyze the sources of the observed NH₃
25 flux emissions of the two different (remote and anthropogenic) deciduous forests, we used the
26 biophysical bi-directional surface model SURFATM-NH₃ in combination with a
27 physiologically based leaf photosynthesis-stomatal conductance model (Collatz et al. 1991)
28 for simulating the NH₃ and CO₂ fluxes in different phenological stages. The good agreement
29 for the energy and *NEE* fluxes between measurements and simulations gives confidence in the
30 model representation of the physical and physiological processes that are important for
31 simulating and analyzing the observed forest - atmosphere NH₃ exchange.

1 **4.1 Forest – atmosphere NH₃ fluxes**

2 **4.1.1 MMSF – a natural green forest canopy**

3 Overall, the daytime magnitude of the NH₃ fluxes from the green canopy at MMSF (up to 50
4 ng NH₃-N m⁻² s⁻¹) and the diurnal pattern of the NH₃ fluxes from the forest were simulated
5 moderately well with SURFATM-NH₃ ($R^2 = 0.45$). In particular, daytime stomatal NH₃
6 emissions are well simulated (in the range 36 and 46 ng NH₃ m⁻² s⁻¹), however slightly
7 overestimated during midday, whereas measured nighttime NH₃ emissions (up to 30 ng NH₃-
8 N m⁻² s⁻¹) were not represented by the model (Figure 5a). Nighttime emissions of NH₃ are
9 rarely reported in the scientific literature because deposition fluxes exceed emission rates in
10 most studies. Exceptions are crop fields and managed grasslands where fertilization causes
11 NH₃ volatilization from soil and fertilizers during both day and night (e.g., David et al. 2009;
12 Sutton et al. 2009). In contrast, the MMSF station represents a remote natural site with very
13 low atmospheric NH₃ concentrations ($\approx 0.5 \mu\text{g NH}_3\text{-N m}^{-3}$) and inferior NH₃ deposition
14 (Figure 6a). If the atmospheric NH₃ concentration is lower than the NH₃ compensation point,
15 natural ecosystems may act as a source of NH₃ (Langford and Fehsenfeld 1992). Sites with
16 low N supply are generally expected to have low NH₃ compensation points (e.g., Massad et
17 al. 2010a; Zhang et al. 2010). The sources of NH₃ emissions are further discussed on the basis
18 of observed and modelled NH₃ emissions in section 4.2.

19 **4.1.2 DK-Sor - a senescent forest influenced by anthropogenic NH₃** 20 **depositions**

21 For DK-Sor, NH₃ depositions up to 250 ng NH₃-N m⁻² s⁻¹ were measured during the first five
22 days when PAI was $\sim 3 \text{ m}^2 \text{ m}^{-2}$ (Figure 5b). The model was not able to represent these
23 deposition rates. Indeed, the measurements exceed the maximum possible flux permitted by
24 turbulent transfer ($F_{\text{max}} = -c_{\text{NH}_3}/r_a$) in this period, as discussed in Hansen et al. (2013),
25 however this simple analysis assumes horizontal and vertical homogeneity and no chemical
26 reactions within the gradient. Following these days, emissions of up to 670 ng NH₃-N m⁻³
27 were observed during the leaf fall period. The emission events during DOY 306–308 and
28 316–318 are well simulated by SURFATM-NH₃ using PAI to scale the influence of litter on
29 the ground layer emission potential. Modelled emissions were strongly controlled by
30 turbulence assessed by the friction velocity. However, the emission fluxes measured during
31 DOY 314–316 are not captured by the model. During these days, the air temperature

1 decreased to below 5°C, and on DOY 316 it increased to above 5°C. The low temperatures
2 are limiting the modelled emissions from the ground layer, as the compensation point depends
3 strictly and exponentially on temperatures (Husted and Schjoerring 1996; Mattsson et al.
4 1997).

5 **4.2 Sources of forest - atmosphere NH₃ fluxes**

6 Simulated forest component NH₃ fluxes (Figure 6a) show that, for MMSF, NH₃ emissions up
7 to 50 ng NH₃-N m⁻² s⁻¹ dominate the daytime net flux due to stomatal release of NH₃ from the
8 leaves, whereas the contribution of simulated soil emissions is insignificant. SURFATM-NH₃
9 simulates very little cuticular absorption during night and morning, but relatively high
10 observed NH₃ emissions at night suggest that cuticular desorption is more important (section
11 4.1) and responsible for nighttime emissions up to 30 ng NH₃-N m⁻² s⁻¹ for this dense natural
12 forest ecosystem.

13 For DK-Sor, being a small forest surrounded by intensively cultivated crop fields, emissions
14 up to 150 ng NH₃-N m⁻² s⁻¹ were observed in the leaf fall period corresponding to
15 approximately 130 % of the net flux at midday. During/after leaf fall, the ground layer
16 contributes almost solely to the modelled NH₃ emissions (Figure 6b). Stomatal NH₃ fluxes are
17 insignificant in the leaf fall period but the average cuticular absorption amount to 30 ng NH₃-
18 N m⁻² s⁻¹. Due to the cold and humid weather (Table 1), modelled cuticular deposition is
19 nearly constant with no diel variation.

20 Less knowledge exists about soil and litter emissions of NH₃ in (semi-)natural ecosystems.
21 Both emission pathways depend strongly on the seasonal variation in canopy physiological
22 functioning and the building of a leaf litter layer on the forest floor that potentially contributes
23 as a source for NH₃ emissions.

24 **4.2.1 Ground layer NH₃ emissions**

25 Walker et al. (2008) measured the soil NH₃ emission potential of a forest exposed to large
26 NH₃ deposition and found it to be 20 (*n* = 34) at a depth of 5 cm, however other studies
27 indicate much higher emission potential of decomposing litter layers (e.g., Zhang et al. 2010).
28 For instance, Wang et al. (2011) observed higher emission potential of (newly) fallen leaves
29 ($\Gamma_g = 300$) compared to senescing leaves ($\Gamma_s = 200$) at the DK-Sor site. Since it may take 1.3-2
30 years before forest leaf litter is totally decomposed (Muller 2003), the seasonal development

1 of forest floor Γ_g is however not known. Any ground layer NH_3 emissions may be absorbed
2 by the overlying leaf layers of closed canopies (Nemitz et al. 2001; Personne et al. 2009). In
3 case of much higher estimates for Γ_g (6000-30000) than measured for the DK-Sor forest ($\Gamma_g =$
4 300) by Wang et al. (2011), the modelled nighttime forest NH_3 emissions of MMSF would be
5 sensitive to the emission potential of the ground layer (Figure 7c). Using such high estimates
6 for Γ_g , the simulated NH_3 emissions of MMSF (Figure 7c) would however exceed the daytime
7 NH_3 flux observations considerably (Figure 5a). Thus, the observed nighttime NH_3 fluxes of
8 the green forest canopy at MMSF are rather caused by foliar emissions or related to
9 transitions in the gaseous-aerosol phases of atmospheric NH_3 not included in the model.

10 4.2.2 Foliar NH_3 emissions

11 Foliar emissions during daytime are very sensitive to stomatal emission potential and stomatal
12 conductance (Figure 7a). In this study, Γ_s was set to 400 to represent a mid-season green
13 forest canopy, following leaf measurements at DK-Sor (Wang et al. 2011). The use of similar
14 parameter value for Γ_s at MMSF and DK-Sor is supported by nearly similar leaf nitrogen
15 concentration of the two sites (Table 1). During nights, this emission source diminishes due to
16 stomatal closure, however a number of recent gas exchange studies suggested that simulated
17 stomatal conductance may be underestimated at night (e.g., Charusombat et al. 2010; Wu et
18 al. 2011), and a significant loss of water through stomata can take place that may not be
19 measured by eddy covariance systems due to low turbulence at night (Caird et al. 2007;
20 Dawson et al. 2007; Fisher et al. 2007). Measurements of LE at MMSF do not indicate
21 considerable nighttime transpiration, e.g., average nighttime LE varies from -2.5 to 10 W m^{-2}
22 in the study period. Nevertheless, similar rates of eddy-covariance nighttime LE measured in
23 Californian AmeriFlux sites were found to significantly underestimate nighttime transpiration
24 as a percent of daily total when compared to sapflow-based analyses for oak-savannah
25 (underestimation by 12 %) and *Pinus Ponderosa* (underestimation by 20 %) (Fisher et al.
26 2007). Even though the simulated nighttime stomatal conductance, transpiration and stomatal
27 NH_3 flux may be underestimated in this study, the large proportion of observed nighttime
28 relative to daytime NH_3 emission flux at MMSF (Figure 5a) suggest that other processes are
29 also involved at night. Assuming no (or low) nighttime stomatal NH_3 emissions in this case
30 (and no significant soil emissions – see 4.2.1), cuticular desorption could be responsible for
31 the observed nighttime NH_3 emissions. This process is not represented in SURFATM- NH_3
32 that was earlier applied for modelling NH_3 fluxes of agricultural sites where cuticular

1 adsorption is the dominant process. Emission of NH_3 from cuticles requires low cuticular
2 resistance and that NH_3 concentrations at the leaf surface exceed those in the surrounding air.
3 Sources of higher leaf surface NH_3 concentrations may occur from stomatal NH_3 emissions or
4 from deposited aerosols that are converted to gaseous NH_3 at the leaf surface.

5 4.2.1 Leaf surface wetness and dew formation

6 The high solubility of NH_3 in water causes leaf surface wetness to be very important for the
7 estimation of NH_3 fluxes. During night, radiative cooling reduces temperature, increases
8 relative air humidity and causes dew formation. Leaf wetness caused by morning dew was
9 found to increase the NH_3 deposition (e.g., Burkhardt et al. 2009), and there is recent evidence
10 that dew can work as a nighttime NH_3 reservoir which is released back to the atmosphere
11 during early morning dew evaporation (Wentworth et al. 2016). Dew formation starts when
12 100% relative air humidity (*RH*) is reached at the actual leaf surface which normally
13 corresponds to about 90 % *RH* of the surrounding air (Burkhardt and Hunsche 2013). At
14 MMSF, nighttime *RH* approaches 100 % following the rain event at night (at 1 h) on DOY
15 251, and *RH* increases to above 90 % throughout the following 3 nights. The largest nighttime
16 NH_3 emissions are however seen on DOY 248–251 (Figure 5a) where *RH* is lower, e.g., it
17 reaches maxima of 75 %, 70 % and 85 % on DOY 248–251.

18 In addition to morning dew formation, leaf wetness can be caused by microscopic (invisible)
19 water films that are formed by deliquescence of hygroscopic leaf surface particles at high *RH*
20 or as a result of transpiration (Burkhardt and Hunsche 2013). For instance, measurements of
21 leaf surface wetness on potato over five days clearly showed two diel peaks with one leaf
22 wetness peak being related to midday transpiration and the other leaf wetness peak being
23 related to increasing *RH* at night (Burkhardt and Hunsche 2013). It is striking that this
24 observed microscopic leaf wetness pattern resembles a bimodal diel curve also observed in
25 the measured NH_3 emissions at MMSF (Figure 5a). Several studies have indicated that such
26 microscopic water films on leaf surfaces may also enhance the emission of NH_3 depending on
27 the concentration of dissolved ions (Sutton et al. 1998; Sutton et al. 2009; Burkhardt and
28 Hunsche 2013; Wentworth et al. 2016). Unfortunately, we do not know how the ammonium
29 concentrations vary overnight, however high ammonium concentration of microscopic leaf
30 water could explain the observed nocturnal NH_3 emissions at MMSF. Theoretically, the
31 deliquescence of aerosols happens when *RH* reaches the deliquescent relative humidity (*DRH*)
32 which is e.g. 62 % for NH_4NO_3 particles and 80 % for $(\text{NH}_4)_2\text{SO}_4$ particles (at 298 K) (Hu et

1 al. 2011). At *DRH*, the solid particles are transformed to larger aqueous solutions
2 (microscopic droplets or water films) with high ion concentration. Hu et al. (2011) measured
3 the hygroscopic growth curve for different ammonium salt particles in laboratory, and they
4 observed gradually decreasing particle size of highly volatile particles such as NH_4NO_3
5 aerosols in response to increasing *RH* (below *DRH*). This led to the suggestion that small (<
6 50 nm) volatile ammonia particles such as NH_4NO_3 aerosols evaporate during the *RH*
7 increasing process whereas this was not observed for the less volatile $(\text{NH}_4)_2\text{SO}_4$ particles (Hu
8 et al. 2011). With increasing *RH* above *DRH*, the saturated solution droplets grow due to
9 additional water condensation onto the salt solution (Hu et al. 2011), however the growth rate
10 was less than expected for very small (< 50 nm) volatile NH_4NO_3 particles. When *RH* reaches
11 100 %, dew is formed. Burkhardt and Hunsche (2013) hypothesized that microscopic leaf
12 wetness occurs on almost any plant worldwide, often permanently, and that it significantly
13 influences the leaf surface-air exchange processes. Further studies are needed to investigate
14 the role of *RH* on the deliquescence of deposited volatile ammonia particles and the likely
15 occurrence of highly concentrated solutions on leaf surfaces.

16

17 4.2.2 Gaseous - aerosol phase interactions

18 The condensation nuclei for microscopic water typically result from deposited aerosols and
19 may form highly concentrated solutions (Burkhardt and Hunsche 2013). Aerosol
20 concentrations of NH_4^+ previously measured at MMSF indicated that the gaseous and aerosol
21 phase concentrations of $\text{NH}_3/\text{NH}_4^+$ were of similar magnitude, but that the aerosol phase
22 typically dominated (Hansen et al. 2015). Measured HNO_3 fluxes at MMSF showed
23 deposition during daytime and emission during night (Hansen et al. 2015). This leads to the
24 suggestion that the apparent nighttime HNO_3 and NH_3 emissions are caused by dissociation of
25 aerosol NH_4NO_3 at (or near) the cool and humid leaf surfaces, and that this source of NH_3
26 could be responsible for the observed nighttime NH_3 emissions at MMSF. This proposed
27 mechanism may also be related to the suggestion that increasing *RH* alters the chemical
28 equilibrium and accelerates the evaporation of very small volatile particles such as NH_4NO_3
29 aerosols (Hu et al. 2011). It was also earlier suggested by Pryor et al. (2011) that NH_3
30 emission fluxes at MMSF could be caused by NH_4NO_3 aerosol evaporation.

1 4.2.3 Vapor pressure deficit and relative humidity

2 In the current study, we used measurement-based values for simulating ground layer and
3 stomatal NH_3 emissions (measured at DK-Sor), and we find that these cannot explain the
4 observed nighttime NH_3 emissions. However, the vapor pressure deficit, VPD (Pa) is found to
5 be strong positively correlated with both the average nighttime LE ($R^2 = 0.82$) and the average
6 nighttime NH_3 flux ($R^2 = 0.90$). The average nighttime LE and F_{NH_3} are also strongly
7 correlated ($R^2 = 0.88$) whereas H and F_{NH_3} are not correlated ($R^2 = 0.09$). These relations
8 suggest a strong association between VPD , LE and F_{NH_3} at night that may be related to VPD -
9 driven transpiration and foliar NH_3 emissions being supported by high ammonium
10 concentrations at the leaf surface due to the formation of microscopic leaf wetness (by
11 transpiration) and deliquescence of ammonia particles. In the case that NH_3 loss to the air
12 occurs by e.g. NH_4NO_3 aerosol deliquescence and evaporation at night, the observed
13 nighttime NH_3 emission would be strongly affected by in-canopy and leaf surface chemical
14 reactions that are excluded in our model. In SURFATM- NH_3 , the leaf surface NH_3
15 concentration is assumed to be zero. For further analysis, the development and application of
16 more advanced physical and chemical models is required to represent microscopic leaf
17 wetness (Burkhardt et al. 2013) and leaf surface NH_3 concentrations that are in equilibrium
18 with the dissolved NH_3 concentrations at the leaf surface (Wichink Kruit et al. 2010) in order
19 to simulate and analyze the importance of bi-directional cuticular NH_3 gaseous exchange.
20 Model representation of inorganic chemistry interactions on the leaf surface requires many
21 input parameters and excessive computation time, and the development of simpler empirical
22 approaches are also needed for application in atmospheric transport models (Massad et al.
23 2010a; Wichink Kruit et al. 2010).

24 **4.3 Seasonal development of the bulk NH_3 emission potentials Γ_s and Γ_g**

25 A number of bi-directional NH_3 exchange models have been developed (e.g., Flechard et al.
26 2013; Nemitz et al. 2000b; Massad et al. 2010a; Sutton et al. 1995; Sutton and Fowler 1993;
27 Wichink Kruit et al. 2010). The parameterization of seasonal dynamics during the growing
28 season to estimate canopy NH_3 compensation point are often vastly simplified (Simpson et al.
29 2012), however new parameterizations are being developed for application to atmospheric
30 transport models (Wichink Kruit et al. 2010). In practice, ecosystem (soil-vegetation) N and
31 NH_4^+ pools are ever changing and Γ may undergo diel, seasonal and annual cycles. Modelling

1 approaches dealing with temperature response of emission potentials should therefore
2 theoretically also deal with temporal Γ dynamics in the various parts of an ecosystem
3 (Flechard et al. 2013; Massad et al. 2010b; Wichink Kruit et al. 2010). Satellite observations
4 have identified deforestation as an important source to atmospheric NH_3 (e.g., van Damme et
5 al. 2015) and that the NH_3 emissions from large scale forest fires are important to include in
6 atmospheric models (van Damme et al. 2014a). All these findings suggest that forest regions
7 should be dynamically included in atmospheric models by taking into account governing
8 processes in relation to both deposition and emission of NH_3 .

9 Ammonia emission potentials of the ground layer (Γ_g) and stomata (Γ_s) are crucial input
10 parameters in bi-directional NH_3 exchange models, in order to simulate the NH_3
11 compensation points of the ground layer and the stomata, respectively. However, realistic
12 measurements of those are difficult and demanding to obtain and only few data exists. Wang
13 et al. (2013) measured the pH and NH_4^+ concentration in the leaf apoplastic solution for the
14 DK-Sor and two coniferous forest sites in order to study the seasonal variation in stomatal
15 NH_3 compensation point. For DK-Sor, they found for the years 2008 and 2009, i.e., very close
16 to the investigated period in this study, that χ_s peaked in the early-season during leaf
17 expansion ($6.8 \mu\text{g NH}_3 \text{ m}^{-3}$) and again in the late-season during leaf senescence ($5.2 \mu\text{g NH}_3$
18 m^{-3}) while in the mid-season, χ_s was lower (around $2.1 \mu\text{g NH}_3 \text{ m}^{-3}$). During leaf senescence
19 N is translocated from the leaves into other parts of the trees leading to lower N
20 concentrations decreasing tissue NH_4^+ concentrations especially in leaves from the canopy top
21 (Wang et al. 2013). This pattern followed the variation in χ_s determined in parallel on the
22 basis of the gas exchange measurements. Because of the difficulty to measure apoplastic
23 concentrations, the significant correlation between Γ_s and the more easily measurable total
24 foliar $[\text{NH}_4^+]$ (Loubet et al. 2002; Mattsson et al. 2009; Wang et al. 2011) can be used to
25 estimate Γ_s . In this study we used the seasonal measurements of Wang et al. (2011) to set $\Gamma_s =$
26 400 for representing a green forest canopy (MMSF) and $\Gamma_s = 200$ to represent a senescing
27 forest canopy (DK-Sor). Apart from observed seasonal variations in Γ_s , differences in leaf N
28 status of the two forests (2.2% for MMSF and 2.5% for DK-Sor0) support the use of different
29 Γ_s values. To improve model simulations, this parameter should be measured for each site and
30 maybe even parameterized with seasonal variation.

31 The other source for NH_3 is the ground layer (Γ_g) and particularly the decomposing litter that
32 we described with the parameter $\Delta LPAI$. Mattsson et al. (2009) showed that the emission

1 potential of litter could be up to 45-60 times higher than for green leaves and stems of an
2 intensively managed grass land. Here we chose a $\Gamma_{g,\min}$ value of 300, corresponding to
3 measurements of newly fallen leaf litter at DK-Sor, and $\Gamma_{g,\max}$ was estimated by trial and
4 error method ($\Gamma_{g,\max} = 18000$). An emission potential of 18000 is high, however, in the range
5 of values found for senescing plant material (e.g., Sutton et al. 2009; Zhang et al. 2010).

6

7 **5 Conclusions**

8 Simulations with the SURFATM-NH₃ model in combination with a canopy stomatal model
9 show that the atmospheric NH₃ flux above a natural green forest canopy (MMSF) was
10 dominated by the stomatal exchange (up to 50 ng NH₃-N m⁻² s⁻¹). For a senescent canopy
11 influenced by anthropogenic NH₃ depositions in the growing season (DK-Sor), the ground
12 layer, or more specifically the fresh decomposing litter layer, was the largest contributor (up
13 to 150 ng NH₃-N m⁻² s⁻¹) to the total simulated NH₃ emissions from the forest to the
14 atmosphere. The measured day-time pattern of the NH₃ flux for MMSF indicates a strong
15 stomatal control by environmental factors with a strong diel signal (radiation, temperature,
16 humidity, CO₂). However, the model underestimated the observed nighttime emissions of
17 NH₃ for the green forest canopy (MMSF). We hypothesize that cuticular desorption is
18 responsible for these observed NH₃ emissions at night. Recent studies found that nights with
19 high *RH* caused morning evaporation of dew to be an important NH₃ source (Wentworth et al.
20 2016). In our study, nighttime NH₃ emissions were observed at nights without dew formation
21 (e.g., higher *VPD*) and may be related to foliar NH₃ emissions induced by a combination of
22 nighttime transpiration and deliquescence of aerosols leading to high ion concentrations of
23 microscopic leaf surface water (Burkhardt and Hunsche 2013). Emissions of NH₃ due to
24 microscopic leaf wetness have to our knowledge not been observed before, however, it seems
25 likely that in-canopy gaseous-aerosol interactions may cause the formation and evaporation of
26 aqueous aerosols and in particular microscopic leaf water at the cool and humid leaf surfaces.
27 Atmospheric NH₃ concentrations at MMSF are consistently low which suggests that cuticular
28 desorption may also take place during daytime where transpired water condenses on leaf
29 surfaces (Burkhardt et al. 1999). However, further investigations including detailed process
30 based measurements and the modelling of bidirectional cuticular NH₃ fluxes are needed in
31 order to obtain more knowledge on this topic.

1 The NH_3 fluxes measured in DK-Sor showed a less pronounced diurnal pattern, however, a
2 pattern where the flux turns from depositions to emissions parallel to the decreasing PAI and
3 an increasing depth of the leaf litter layer on the forest floor comprising an important NH_3
4 source during leaf fall for deciduous forests. The model was not able to represent the
5 deposition rates before leaf fall but simulated well the emission events following on leaf fall
6 using PAI to scale the influence of litter on the ground layer emission potential. We conclude
7 from this study that deciduous forests potentially comprise a natural source of NH_3 to the
8 atmosphere, and that it is crucial to take into account the bi-directional exchange processes
9 related to both the stomatal, cuticular and ground layer pathways in order to realistically
10 simulate natural forest–atmosphere fluxes of NH_3 . We conclude that the combination of flux
11 measurements and modelling is a robust approach in order to understand the important,
12 however difficult to measure all relevant processes and parameters of the NH_3 exchange with
13 the atmosphere. More specialized studies of measurement campaigns measuring particularly
14 the bulk ground layer emissions potential (I_g) as well as the potentials for the two individual
15 ground layer contributors; the soil and the litter layer, are needed in order to obtain improved
16 model parameterizations.

17

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1

2 **Appendix A**

3 **Description of the two-layer bi-directional NH₃ model**

4 The two-layer bi-directional NH₃ exchange model simulates the total net atmospheric NH₃
5 flux (F_T ; ng m⁻² s⁻¹) as a sum of each of the forest component fluxes; the stomatal NH₃ flux
6 (F_s ; ng m⁻² s⁻¹), the cuticular NH₃ flux (F_w ; ng m⁻² s⁻¹) and the ground NH₃ flux (F_g ; ng m⁻² s⁻¹)
7 which are all related to the NH₃ canopy compensation point (χ_c ; ng m⁻³) (Nemitz et al.
8 2000b):

$$9 \quad \chi_c = \frac{\chi_a(r_a r_b)^{-1} + \chi_s(r_a r_s)^{-1} + (r_b r_s)^{-1} + (r_g r_s)^{-1} + \chi_g(r_b r_g)^{-1}}{(r_a r_b)^{-1} + (r_a r_s)^{-1} + (r_a r_w)^{-1} + (r_b r_g)^{-1} + (r_b r_s)^{-1} + (r_b r_w)^{-1} + (r_g r_s)^{-1} + (r_g r_w)^{-1}} \quad (\text{A1})$$

$$10 \quad F_s = -\frac{\chi_c \chi_s}{r_s} \quad (\text{A2})$$

$$11 \quad F_w = -\frac{\chi_c}{r_w} \quad (\text{A3})$$

$$12 \quad F_g = -\frac{\chi_c \chi_g}{r_g} \quad (\text{A4})$$

13 where r_a (s m⁻¹) is the aerodynamic resistance of the canopy, r_b (s m⁻¹) is the quasi-laminar
14 boundary layer resistance of the canopy, r_s (s m⁻¹) is the stomatal resistance, r_w (s m⁻¹) is the
15 cuticular resistance, r_g (s m⁻¹) is the ground layer resistance which represents a series of the
16 in-canopy aerodynamic resistance and the quasi laminar boundary-layer resistance of the
17 ground layer (Nemitz et al. 2001). χ_g and χ_s (ng m⁻³) are the ground layer compensation point
18 and the stomatal NH₃ compensation points, respectively.

19

20 **Appendix B**

21 **Description of the coupled photosynthesis-stomatal conductance model**

22 The coupled leaf photosynthesis and stomatal resistance model (Collatz et al. 1991) was
23 parameterized for the specific forest sites using measured plant area index, PAI (m² m⁻²) as a
24 proxy of the leaf area index (LAI). The model uses input data of the air temperature, T_a (°C),
25 soil temperature, T_s (°C), leaf temperature, T_L (°C), relative humidity, RH (%), photosynthetic
26 active radiation, PAR (W m⁻²), sensible heat flux, H (W m⁻²), wind speed, u (m s⁻¹), friction

1 velocity, u^* (m s^{-1}), and PAI ($\text{m}^2 \text{m}^{-2}$), to simulate the stomatal conductance, g_s (m s^{-1}) and the
 2 leaf photosynthesis (or net carbon assimilation), A_n ($\mu\text{mol m}^{-2} \text{s}^{-1}$), in an iterative setup (Figure
 3 9). A_n is estimated as the minimum of three potential capacities and the leaf dark respiration,
 4 R_D ($\mu\text{mol m}^{-2} \text{s}^{-1}$), (Collatz et al. 1991, Farquhar et al. 1980b) :

$$5 \quad A_n = \min \{J_E, J_C, J_S\} - R_D \quad (\text{B1})$$

6 J_E is the light-limited assimilation rate ($\mu\text{mol m}^{-2} \text{s}^{-1}$), J_C is the rubisco-limited assimilation
 7 rate ($\mu\text{mol m}^{-2} \text{s}^{-1}$), and J_S is the assimilation rate due to the limitation of the export of
 8 assimilates inside the leaf ($\mu\text{mol m}^{-2} \text{s}^{-1}$) simulated as (Collatz et al. 1991):

$$9 \quad J_E = a \alpha Q \frac{C_i - \Gamma^*}{C_i + 2\Gamma^*} \quad (\text{B2})$$

$$10 \quad J_C = \frac{V_{c,\max} (C_i - \Gamma^*)}{C_i + K_c \left(1 + \frac{O_2}{K_o}\right)} \quad (\text{B3})$$

$$11 \quad J_S = 0.5V_{c,\max} \quad (\text{B4})$$

12 where a is the leaf absorptivity of PAR , α ($\mu\text{mol m}^{-2} \text{s}^{-1}$) is the maximum quantum yield,
 13 Q is PAR , C_i (Pa) is the internal CO_2 pressure, Γ^* (Pa) is the CO_2 compensation point, $V_{c,\max}$
 14 ($\mu\text{mol m}^{-2} \text{s}^{-1}$) is the maximum carboxylation rate of Rubisco, O_2 is the oxygen intercellular
 15 partial pressure (Pa), and K_c (40.4 Pa) and K_o (24,800 Pa) are the Michaelis constant for CO_2
 16 fixation and oxygen inhibition, respectively. R_D can experimentally be determined by gas
 17 exchange measurements of leaves, but here it is determined by a fraction of $V_{c,\max}$ following
 18 Collatz et al. (1991):

$$19 \quad R_D = 0.015V_{c,\max} \quad (\text{B5})$$

20 The stomatal conductance is simulated following Ball et al. (1987):

$$21 \quad g_s = m \frac{A_n h_s}{C_s} + b \quad (\text{B6})$$

22 where h_s (%) is the relative humidity at the leaf surface, C_s (Pa) is the CO_2 partial pressure on
 23 the leaf surface, and $m = 7$ and $b = 0.01 \text{ mol m}^{-2} \text{s}^{-1}$ are constants.

24 The leaf temperature, T_L ($^\circ\text{C}$), is simulated as:

$$25 \quad T_L = \frac{H r_{\text{abh}}}{\rho c_p} + T_a \quad (\text{B7})$$

26 where r_{abh} (s m^{-1}) is the total resistance to heat, ρ (kg m^{-3}) is the air density, and c_p is the
 27 specific heat for air at constant pressure ($\text{J kg}^{-1} \text{K}^{-1}$).

1 $V_{c,max}$ is strongly dependent on the availability of leaf nitrogen and leaf temperature. Hence it
 2 is expected that $V_{c,max}$ at 25 ° C is higher for the DK-Sor site than for the MMSF site, since
 3 the DK-Sor site is located in an agricultural region exposed to large NH_3 deposition (Skiba et
 4 al. 2009). The applied parameter values for $V_{c,max}$, at 25 ° C for DK-Sor ($100 \cdot 10^{-6} \text{ mol m}^{-2} \text{ s}^{-1}$)
 5 and MMSF ($70 \cdot 10^{-6} \text{ mol m}^{-2} \text{ s}^{-1}$) are within the range of values found and used for this plant
 6 functional type (PFT) in other terrestrial biosphere models (Kattge et al. 2009; Rogers 2014).
 7 Three different parameterizations of the soil and ecosystem respiration were furthermore
 8 tested in order to obtain the best representation; two of them based on the temperature-
 9 controlled soil (R_s) or ecosystem (R_{eco}) respiration model from Lloyd and Taylor (1994) and
 10 one total ecosystem respiration (TER) parameterized for DK-Sor by Wu et al. (2012):

$$11 \quad R_s = R_{10} \times \exp\left(308.56 - \left(\frac{1}{T_{ref} - T_0} - \frac{1}{T_s - T_0}\right)\right) \quad (B8)$$

$$12 \quad R_{eco} = R_{eco,ref} \times \exp\left(308.56 - \left(\frac{1}{T_{ref} - T_0} - \frac{1}{T_a - T_0}\right)\right) \quad (B9)$$

$$13 \quad TER = R_{eco,ref} Q_{10}^{\frac{T_s - T_0}{10}} \quad (B10)$$

14 where R_{10} ($\mu\text{mol m}^{-2} \text{ s}^{-1}$) is the soil respiration at 10°C, T_s (°C) is soil temperature, T_{ref} (°C) is
 15 the reference temperature set to 10°C as in the original model, T_0 (°C) is a regression constant
 16 of -46.02°C (Lloyd and Taylor 1994), T_{air} (°C) is the ambient air temperature, $R_{eco,ref}$ ($\mu\text{mol m}^{-2}$
 17 s^{-1}) is the respiration at T_{ref} estimated from nighttime data, and Q_{10} is the temperature
 18 sensitivity parameter and set to a constant value of 2. The parameterization of TER by Wu et
 19 al. (2012) was used for DK-Sor, and the general parameterization (R_{eco}) was used for MMSF.

20 Subtracting the simulated soil respiration (R_s) from A_c (canopy net photosynthesis per ground
 21 area) being A_n (leaf net photosynthesis per leaf area) upscaled to the canopy scale using LAI
 22 as proposed by Sellers et al. (1992), the net CO_2 exchange was calculated at canopy scale
 23 ($NEE = -A_c + R_s$) and can be directly tested with eddy covariance CO_2 flux measurements
 24 (Pilegaard et al. 2011).

25

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1 **Table 1: Site location and characteristics during the measurement periods. Mean including standard deviation is given for**
 2 ***PAI*, temperature and rain.**

		MMSF	DK-Sor
Lat/Lon		39°53'N, 86°25'W	55°29'N, 11°38'E
Forest type		Temperate deciduous, mixed	Temperate deciduous, beech
Canopy height	m	~28	~26
Tree age	years	80 - 90	82
Summer <i>PAI</i>	m ² m ⁻²	4.6	4.6
Soil type		Mesic Typic Dystrochrepts	Alfisols or Mollisols
Leaf N status	%	2.2	2.5
Measurement period			
Dates		5 - 10 September 2013	21 October – 15 November 2010
DOY		248 – 253	294 – 319
Mean <i>PAI</i>	m ² m ⁻²	4.5 ± 0.0	2.2 ± 0.9
Mean Temp.	°C	24.5 ± 3.3	6.7 ± 2.6
Total Rain	mm	12.8	124.0

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1 **Table 2: List of input parameters for the SURFATM-NH₃ model for the MMSF and DK-Sor model setups. Pedotransfer functions applicable to Danish soil types are used (Madsen and**
 2 **Holst 1988) and some soil parameters for the DK-Sor site were measured in the NitroEurope project (2006-2011) and provided from there.**

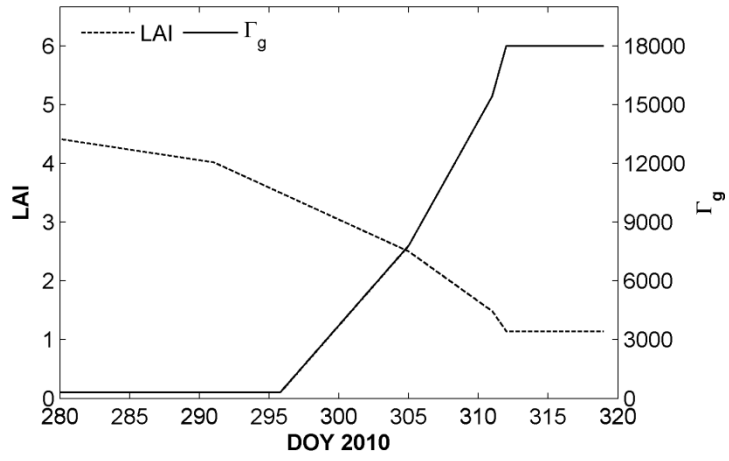
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	Soil parameters	Unit	Range	MMSF/DK-Sor	Source
	Soil depth	m		0.86 / 0.80	Thompson et al. (2011) / NitroEurope IP
	Soil density	kg m ⁻³		1220 / 1038	Measured / NitroEurope IP
	Soil humidity at field capacity	kg (H ₂ O) kg ⁻¹ (soil)	[0.15 – 0.4]	0.36 / 0.36	Estimated / calculated using pedotransfer function
	Soil humidity at wilting point	kg (H ₂ O) kg ⁻¹ (soil)	[0.05 – 0.25]	0.22 / 0.24	Estimated / calculated using pedotransfer function
	Thermal soil conductance (wet)	W m ⁻¹ K ⁻¹	[1.6 – 2.2]	1.60 / 1.00	Monteith and Unsworth (1990)
	Thermal soil conductance (dry)	W m ⁻¹ K ⁻¹	[0.2 – 0.3]	0.28	Monteith and Unsworth (1990)
	Soil porosity	-	[0.25 – 0.5]	0.55	Thompson et al. (2011) / Estimated
	Soil tortuosity parameter	-	[2 - 4]	2.00	Estimated
	Soil Roughness	m	[0.001 – 0.5]	0.02	Estimated
Chemical constants					
K_H	Henry Constant for NH ₃	-		10 ^{-3.14}	Loubet (2000)
K_d	Dissociation constant for acid-base dissociation NH ₄ ⁺ /NH ₃	mol l ⁻¹		10 ^{-9.25}	Bates and Pinching (1950)
Vegetation parameters					
	Leaf width	m	[0.03 – 0.5]	0.15 / 0.10	Measured
	Canopy height	m		28 / 26	Measured
	Max stomatal conductance	m s ⁻¹		400	Collatz et al. (1991)
	Efficiency coefficient for plant area index			0.25	Estimated
	Radiation attenuation coefficient	-	[0.5 – 0.8]	0.80 / 0.85	Estimated / calculated using radiation data
	Wind attenuation coefficient	-	[1.5 - 5]	2.20	Estimated
Ammonia emission potentials					
$\Gamma_{g,min}$	Min ground layer emission potential	-		300	Estimated
$\Gamma_{g,max}$	Max ground layer emission potential	-		18000	Estimated
Γ_s	Leaves (stomata)	-	[0 - 600]	400 / 200	Estimated / Wang et al. (2011)

1 **Table 3: Model error statistics including the number of valid observations, n , Pearson correlation coefficient, R^2 , root**
 2 **mean squared error, $RMSE$, and the Concordance coefficient, CCC (in $W\ m^{-2}$ for energy fluxes and $\mu g\ NH_3-N\ m^{-2}\ s^{-1}$ for**
 3 **ammonia fluxes). See text (section 3.1.1) and Figure 3 for energy balance closure.**

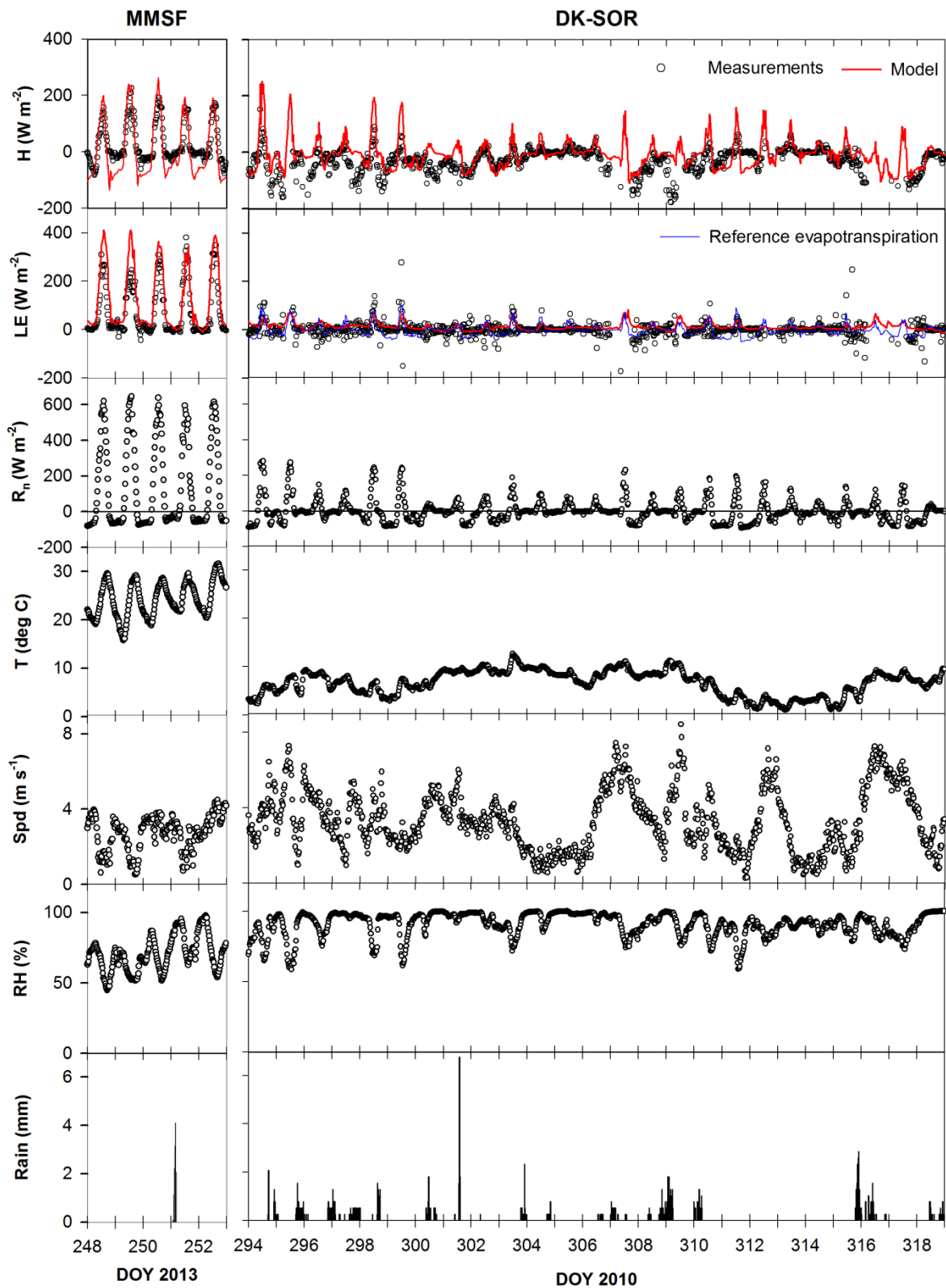
MMSF	n	R^2	$RMSE$	CCC
H	240	0.78	59.17	0.69
LE	240	0.87	71.83	0.78
G	240	0.40	21.46	0.22
F_{NH_3}	209	0.51	14.03	0.43
DK-Sor	n	R^2	$RMSE$	CCC
H	1198	0.17	71.94	0.14
LE	1199	0.07	37.32	0.04
G	1200	0.65	10.97	0.32
F_{NH_3}	1020	0.62	94.29	0.60

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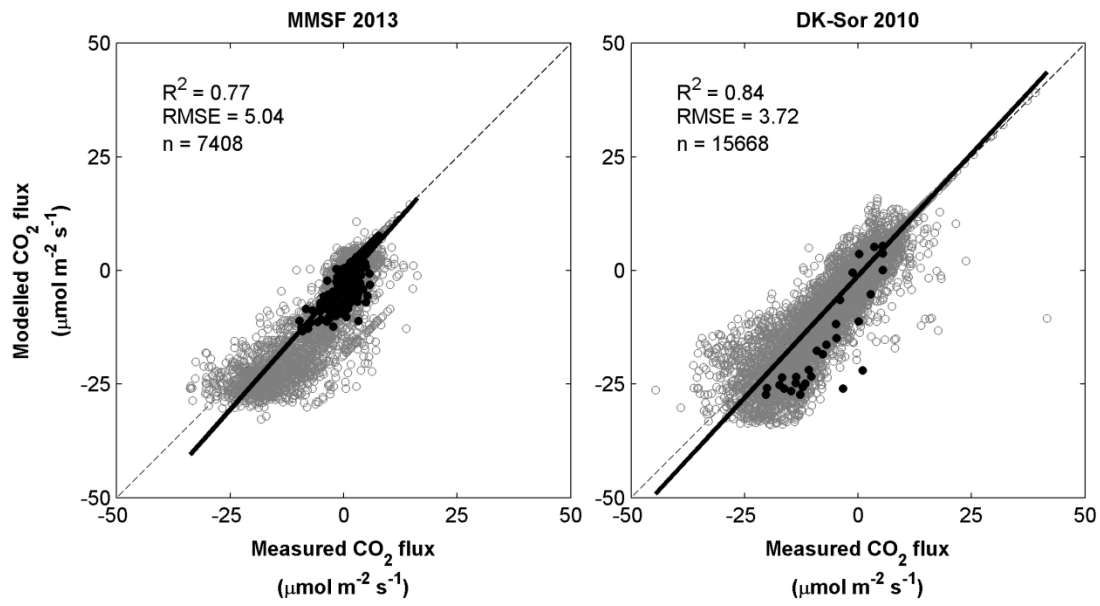
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 2 **Figure 1: Parameterization of the ground layer emission potential $\Gamma_g = [\text{NH}_4^+]/[\text{H}^+]$ with the decrease in ΔLPAI exemplified**
 3 **using *PAI* data from the DK-Sor site.**

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 2 **Figure 2: Comparison of the measured (dots) and the simulated (red lines) energy fluxes ($W m^{-2}$) of sensible heat (H) and**
 3 **latent heat (LE) for the MMSF and DK-Sor sites. For the DK-Sor site the FAO Penman-Monteith reference**
 4 **evapotranspiration is also shown with (blue line). Measurements of net radiation, Rn ($W m^{-2}$), air temperature, T (degree**
 5 **celcius), wind speed, Spd ($m s^{-1}$), relative humidity, RH (%), and $rain$ (mm) are shown in the lowest graphs.**

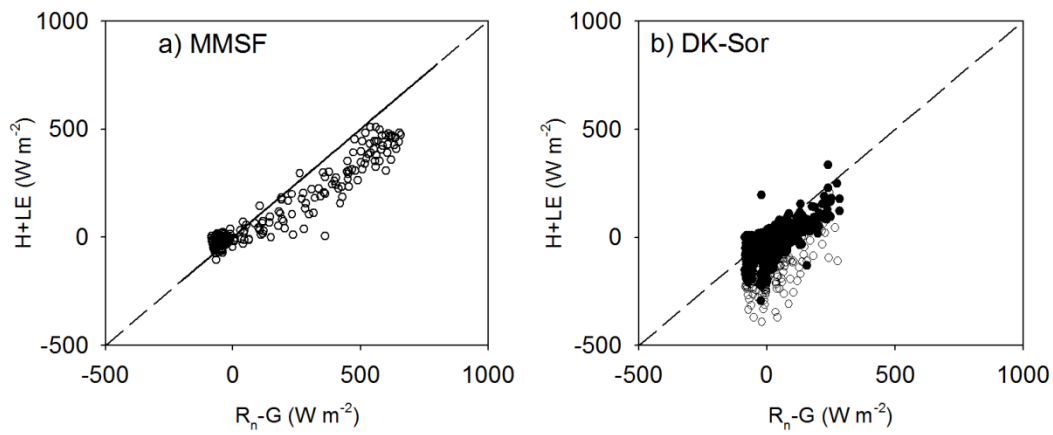
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2 Figure 3: Scatterplot of the modelled vs. the measured CO₂ flux (μmol m⁻² s⁻¹) for the DK-Sor site for the full year of 2010
 3 (left) and for the MMSF site for the full year of 2013 (right). Black filled symbols show the data points during the
 4 measurement periods. *R*² is the coefficient of determination, *RMSE* is the root mean square error, and *n* is the number of
 5 valid sampling points.

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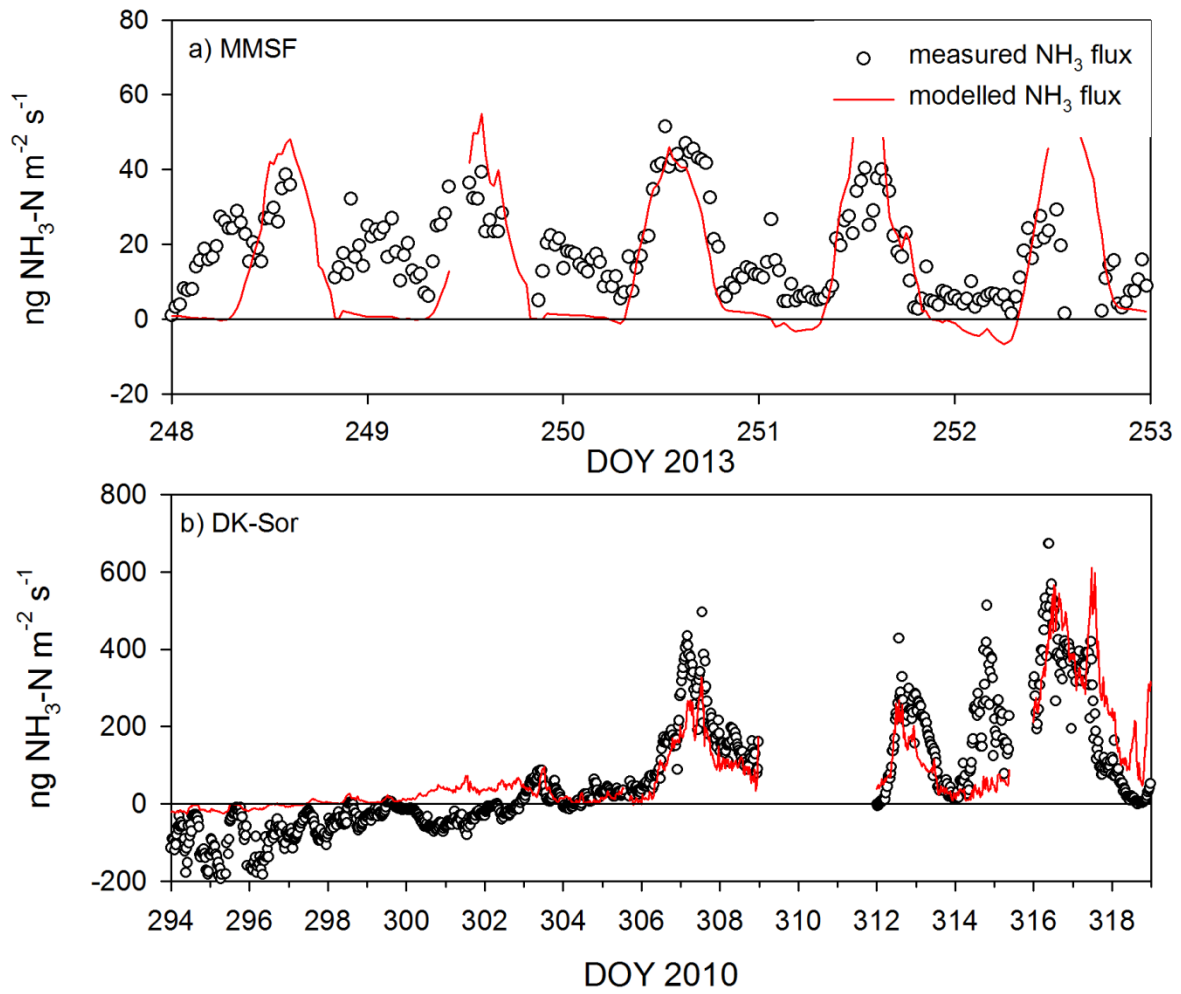
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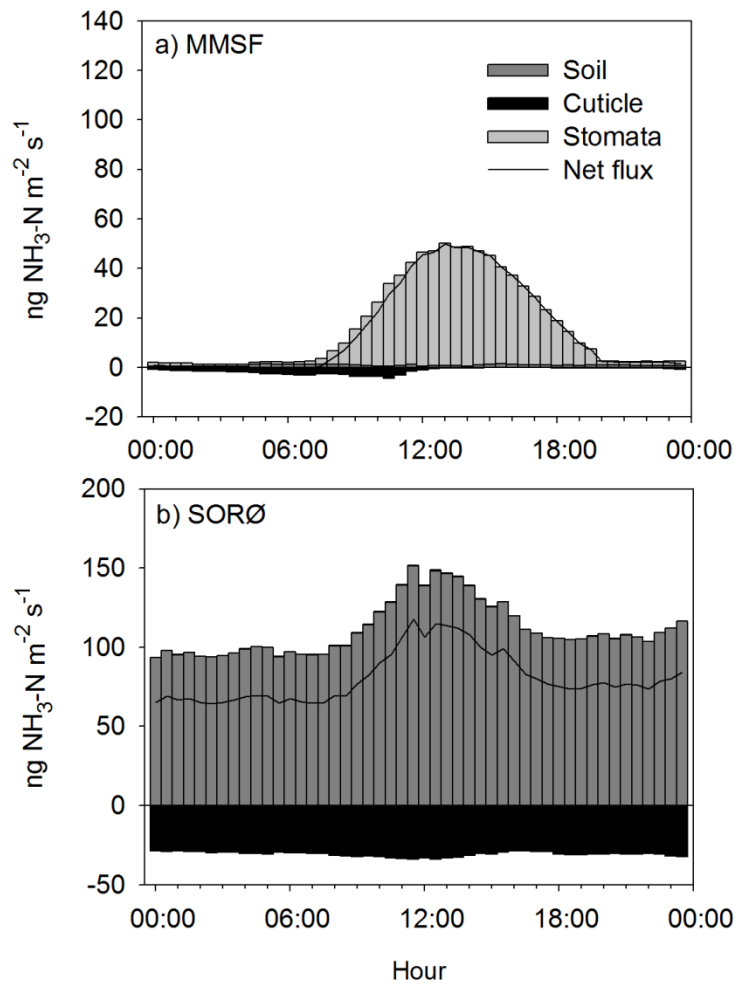
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Figure 4: Scatter plots comparing the measured available energy ($R_n - G$) to the measured turbulent energy fluxes ($H + LE$) for a) the MMSF site and b) the DK-Sor site during the measurement periods. For (b) the filled circles indicate measurements during periods where the wind speed was 0-5 m s^{-1} and the open circles represent data measured at wind speeds higher than 5 m s^{-1} .



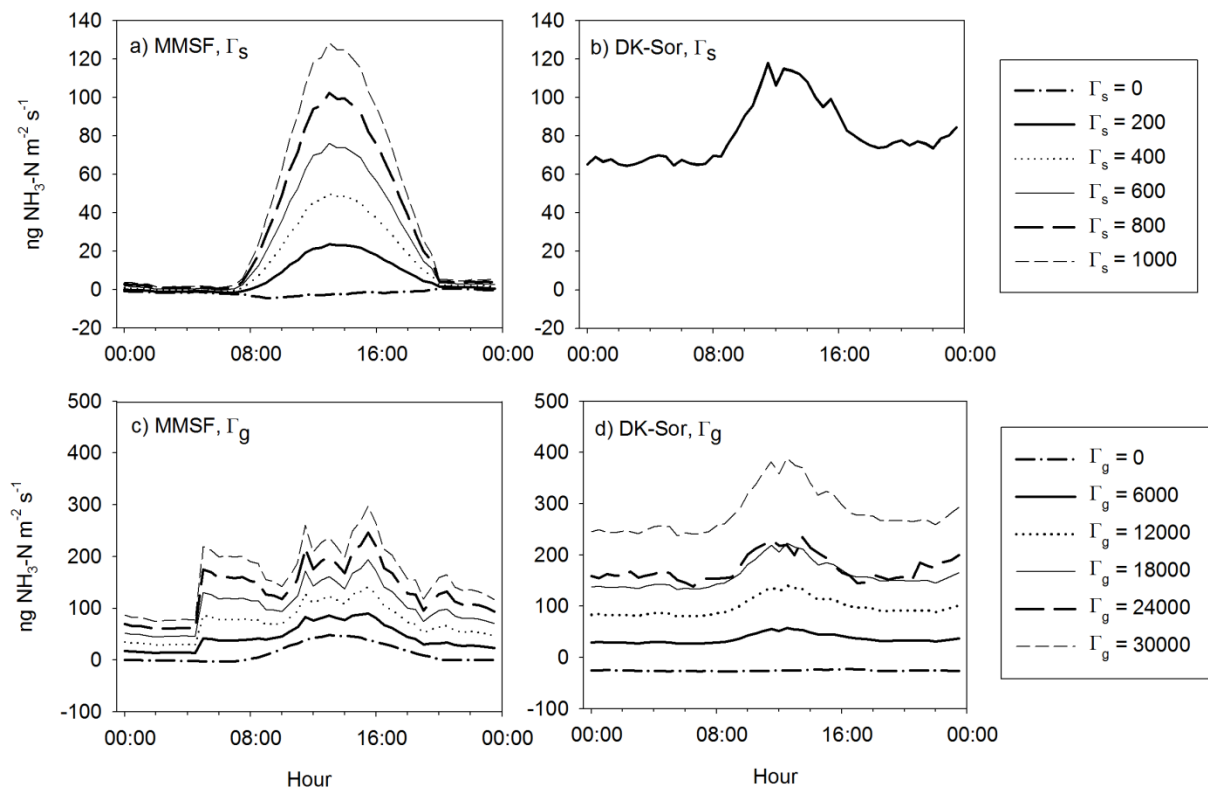
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Figure 5: Comparison of the measured (dots) and the simulated (lines) NH_3 fluxes ($\text{ng NH}_3 \text{ m}^{-2} \text{ s}^{-1}$) for a) the MMSF site and b) the DK-Sor site during the measurement periods.

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Figure 6: Simulated mean forest component fluxes ($\text{ng NH}_3\text{-N m}^{-2} \text{s}^{-1}$) from the ground layer (light grey), cuticles (dark grey), and stomata (white) for a) MMSF site, which represents a green canopy and b) DK-Sor site, which represents the leaf fall period. The solid line shows the mean NH_3 net flux ($\text{ng NH}_3\text{-N m}^{-2} \text{s}^{-1}$).



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Figure 7: The simulated mean NH_3 flux ($\text{ng NH}_3\text{-N m}^{-2} \text{s}^{-1}$) for a green forest canopy (MMSF) and for a leaf fall forest canopy (DK-Sor) with different stomata emission potentials (Γ_s) (a and b) and different ground layer emission potentials (Γ_g) (c and d). In panel b) all results fall on one single line, indicating that the stomatal NH_3 pathway is negligible.