



NH₃ concentrations over Europe – a static and dynamic approach with WRF-Chem

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Spatial, temporal and vertical distribution of ammonia concentrations over Europe – comparing a static and dynamic approach with WRF-Chem

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Abstract

The study focuses on the application of a dynamic ammonia emission into the Weather Research and Forecasting Chemistry model (WRF-Chem) and the influence on the simulated ammonia concentrations and the overall model performance. We have focused on agricultural ammonia sources and have analysed both hourly and daily patterns of ammonia emissions and concentrations at measurement sites located in agricultural areas or influenced by this activity. For selected episodes, we have also investigated the 3-D patterns of the ammonia concentrations in the atmosphere. The application of the dynamic ammonia emission into the WRF-Chem model (the “DYNAMIC” simulation) results in an improvement of the modelled daily ammonia concentrations in comparison to a static approach (the “BASE” simulation), which is currently widely used in chemical transport models. In the case of hourly resolution, we have observed an improvement for the DYNAMIC approach for the winter and autumn seasons, but for the entire year the modelled hourly ammonia peaks are shifted toward the afternoon hours if compared with measurements. This study indicates that the current description of the diurnal cycle of the ammonia concentration from fields is not accurate and more research is needed in order to improve the processes that describe the emission from fertilised fields. The results suggest that the governing processes in relation to the diurnal cycle are the atmospheric mixing and the emission strength. Therefore, an improved description of the diurnal profile of ammonia concentrations within atmospheric models requires a better description of the planetary boundary layer height and a stronger daily pattern of ammonia emission, e.g. through increased evaporation or increased fluxes from the surface.

1 Introduction

Ammonia (NH₃) is the most abundant form of reduced nitrogen in the gas-phase within the atmosphere (Behera et al., 2013). Ammonia contributes to both formation of partic-

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ulate matter (PM) and deposition of reactive nitrogen to the environment (Banzhaf et al., 2013; Hertel et al., 2012; Reis et al., 2009). Ammonia plays a decisive role in particulate matter formation chemistry by determining the amount of ammonium sulphate and ammonium nitrate as PM constituents (Bessagnet et al., 2014). Erisman et al. (2008) estimate that NH₃ emissions from agriculture give a substantial contribution (13 %) to the PM concentration in Europe and thereby adds significantly to the external costs related to air pollution in Europe (Brandt et al., 2013). The deposition of nitrogen in the form of ammonia can result in eutrophication of sensitive ecosystems and to acidification of the soil (Bouwman et al., 1997). An enhanced load of nitrogen in terrestrial ecosystems has been found to correlate with loss of biodiversity (Stevens et al., 2004) and can increase ecosystem vulnerability to extreme weather and insect attacks (Bobink et al., 2010). Despite the importance, there are still many knowledge gaps on ammonia. Long term observational data series are in general scarce compared to e.g. ozone (Tørseth et al., 2012), and hourly observations of ammonia are even more rare due to limitations in existing monitoring networks such as EMEP (Tørseth et al., 2012).

Agriculture was responsible for 94 % of the total NH₃ emissions in the EU in 2010 (European Environment Agency, <http://www.eea.europa.eu/>). Agricultural emissions are related to farm buildings, manure, fertilisers and grazing animals, and are strongly influenced by climate and weather (Skjøth and Geels, 2013). NH₃ emission varies primarily with temperature and air velocity (Monteny and Erisman, 1998). The volatilization potential nearly doubles for every 5 °C and varies significantly through the day and season (Gyldenkerne et al., 2005; Sutton et al., 2013). NH₃ emission is also controlled by water availability, which allows nitrogen compounds to dissolve, be taken up by organisms and be released through decomposition (Sutton et al., 2013). The regional variation reflects local production methods and agricultural practice, which to a large extent is governed by regional scale climate conditions (Skjøth et al., 2011). Despite these well-known dependencies of climate and meteorology on ammonia emissions, the emissions are handled in a very simplified manner in most atmospheric models (Skjøth et al., 2011). Many integrated effects of meteorology and climate on ammonia

remain to be studied and this has been highlighted by IPCC (Kirtman et al., 2013) as an area that is poorly understood. Improvement of representation of processes that lead to ammonia emission in atmospheric models has therefore frequently been highlighted as an area that needs scientific attention (Flechar et al., 2013; Sutton et al., 2013).

Ammonia emission inventories for regional air-quality models in Europe have mainly been based on a bottom-up approach, where activity statistics are combined with standard emission factors to estimate annual emissions. The temporal variation is typically included as a standardised seasonal variation with a monthly profile (Schaap et al., 2005). Recently, Sutton et al. (2013) suggested two long-term goals on ammonia modelling within CTMs. Firstly, the same meteorological data should be used to drive the emission, the chemistry-transport and the bi-directional exchange. Secondly, the emission should be calculated online in the CTM's whereby the feedbacks between the ammonia emissions and climate can be included. Recently, Zhu et al. (2015) indicated that updates to the governing processes on both dynamics and physics concerning NH₃ need improvements. Previous studies have shown significant improvements in model performance by replacing static seasonal variations by a dynamic approach which accounts for physical processes like volatilization of NH₃ (Skjøth et al., 2011). Similarly, other modelling studies have shown that some atmospheric components are sensitive to the formulation of the ammonia emission (Gilliland et al., 2003). This highlights the need for a better understanding of ammonia emissions and how this is implemented in CTM models. Results presented by Werner et al. (2015) suggest that implementing this dynamical approach improves simulations even in areas with limited information about location of the agricultural fields, livestock and agricultural production methods.

Recent studies on modelling atmospheric ammonia with CTMs have focused on the implementation of the bidirectional NH₃ exchange between atmosphere and surface (Bash et al., 2013; Wichink Kruit et al., 2012; Zhu et al., 2015), impact of ammonia emissions on concentrations of secondary inorganic aerosols (Banzhaf et al., 2013; Bessagnet et al., 2014; Hamaoui-Laguel et al., 2014), investigations of the role of different natural emission sources of ammonia (Hansen et al., 2013; Riddick et al., 2014)

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and improvements in the representation of ammonia emission from different agricultural activities, e.g. livestock emission (Zhu et al., 2015) and mineral fertilizers (Hamaoui-Laguel, 2014). Focus on the processes that generate ammonia emissions and the initial dispersion has however had limited attention in the development of existing CTM models. Addressing this knowledge gap is one of the objectives in the FP7 project ECLAIRE (Sutton et al., 2013).

In our study we aim at improving the basic understanding of ammonia in the atmosphere. We will also identify current limitations in relation to the governing processes that cause ammonia emissions and initial dispersion due to meteorological parameters. For this purpose we have implemented the emission from a dynamical ammonia emission model into WRF-Chem and evaluated the model results against a commonly used static approach for describing the emissions. To reveal the limitations in descriptions of the processes, we analyse and compare the model results against hourly observations from one station and daily observations from six stations during the year 2012. The sites have been chosen because they are all located in areas under the influence of agricultural activities. For selected episodes in February, April, July and October, we have analysed both the vertical and temporal development of ammonia concentrations in order to highlight the governing processes that are responsible for initial distribution of ammonia concentrations using meteograms. This analysis of ammonia concentrations both vertically and temporally has not been undertaken before with regional scale atmospheric models. This therefore represents a substantial step forward in understanding the behaviour of ammonia in the atmosphere.

2 Data and methods

2.1 WRF-Chem model

The Weather Research and Forecasting model coupled online with chemistry (WRF-Chem) was used to simulate the meteorological conditions and ammonia concentra-

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ble 2). The seasons were calculated as follows: winter (January, February and December), spring (March, April and May), summer (June, July and August) and autumn (September, October and November). We also explored the temporal variation of ammonia concentrations with height using meteograms for four selected episodes and compared the surface concentrations with the PBL height. Secondly, we calculated statistics for the entire study period and the six measuring sites using the daily mean NH₃ concentrations, both for the BASE and DYNAMIC simulations (Table 3). The following measures were used to summarize the WRF-Chem model performance for both runs: factor of two (FAC2), mean bias (MB), normalized mean bias (NMB), root mean squared error (RMSE) and correlation coefficient (*R*). We then calculated mean statistics from all stations according to the seasons (Table 4). The number of observations available for each season is listed in Tables 2, 3, and 4 as “n”. Finally, we evaluated the spatial pattern in ammonia emission by calculating the day of the year (Julian Day) for which the model estimated the highest hourly ammonia concentrations, with calculation performed independently for each grid cell.

3 Results

The results are presented in the following order: (1) temporal pattern of ammonia emission and concentrations including vertical distribution of concentrations, (2) spatial distribution of NH₃ concentrations for the BASE and DYNAMIC simulations, (3) comparisons of WRF-Chem ammonia concentrations with observations for the both BASE and DYNAMIC runs.

3.1 Temporal and vertical distribution of NH₃ emissions and concentrations

3.1.1 Hourly ammonia emission and concentration – temporal pattern

The hourly profiles of ammonia emissions (DYNAMIC approach only) and modelled and measured concentrations (both BASE and DYNAMIC) have been compared for

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tions are at the surface layer with a linear decrease towards the upper layers. The WRF-Chem model calculates the highest ammonia concentrations at night time, with the time of the maximum varying according to the month. During the periods analysed, this usually occurs before or at midnight in February and July, and after or at midnight in April. In October, the maximum values appear both before and after midnight. In July and October, there are individual days with increased concentrations in the upper layers (4 July and 5 October), which are accompanied by high surface concentrations. The daily pattern of the ammonia concentration is seen to be less regular for some short periods – e.g. on the 5 April, which is related to a precipitation event during that day and washout of ammonia from the atmosphere. For all episodes, ammonia concentration peaks are negatively correlated with the PBL height (Fig. 4). In fact, the periods with the strongest diurnal pattern of ammonia concentrations are also on the days with large differences in PBLH between day and night. For April, we have illustrated the vertical distribution of NO₃⁻, NH₄⁺ and SO₄²⁻ concentrations in the Supplement (Fig. S2). It can be seen that high concentrations of airborne aerosols (Fig. S2), are slightly shifted with a later peak compared to the peak values of ammonia concentrations (aerosol peaks are about 1–2 h later). These figures also show that the maximum aerosol concentrations appear above the surface layer – for NO₃⁻ and NH₄⁺ this is usually about 200 m above ground level.

3.1.3 Daily ammonia concentrations – temporal pattern

The time series with observed and modelled (BASE and DYNAMIC) ammonia concentrations are presented for all stations in Fig. 5. The modelled peak of ammonia concentrations starts at the beginning of February for the BASE simulation and is moved towards March and April for the DYNAMIC simulation. This results in a better agreement with measurements for each station (Fig. 5). However, the magnitude of this spring time peak is overestimated for all the stations, both for BASE and DYNAMIC. The spring peak is much more extended in time in the BASE simulation, if compared to both the observations and the DYNAMIC simulation. The DYNAMIC simulation pro-

vides a second peak in autumn, which is not present in the observed data for Danish sites and is less pronounced for the BASE simulation.

3.2 Spatial distribution of NH₃ concentrations

The BASE simulation shows that the highest ammonia concentrations in February are in western France, northern Italy and several regions of Eastern Europe (Fig. 6). The DYNAMIC simulation shows that the maximum values in February are for the same regions, but the NH₃ air concentrations are 10 times lower if compared to the BASE simulation. In April, high concentrations with value above 10 µg m⁻³ are found in Germany, France, Denmark, and northern Italy for the BASE simulation. For the DYNAMIC simulation high concentrations cover the entire central Europe. In October, both model runs show similar spatial distribution of NH₃ concentrations.

Figure 7 shows the Julian day number (for calendar year 2012), for which the model calculated the highest hourly ammonia concentrations. This was calculated to check whether there is a clear south–north pattern, with increasing number of day towards north for the DYNAMIC simulation and compare this with the results for the BASE simulation. In Central, Eastern and Western Europe, this annual peak day appears later for the DYNAMIC simulation than for the BASE simulation. In the areas with low or no ammonia emissions (e.g. northern Scandinavia) the difference is small between the two simulations. For BASE it is at around day number 60–70 in Central and Eastern Europe, day number 60–90 in Western Europe and above day number 190 in Scandinavian Peninsula. The DYNAMIC simulation shows day number 100–120, day number 70–90, and above day number 190, respectively. Spring application of manure and fertilizers (day number 90–120) described in the DYNAMIC simulation is responsible for the maximum ammonia concentrations in most of the European countries with high ammonia emissions, like e.g. the UK, Denmark, France or Poland.

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3.3 Model evaluation for the BASE and DYNAMIC simulation

Comparison of modelled and measured hourly NH_3 concentrations for Harwell for the BASE simulation indicates the lowest NMGE and the highest R for spring (Table 2). The lowest performance according to NMGE and R is for autumn and summer, respectively. For the DYNAMIC simulation all statistics are improved, in comparison to BASE, for the winter and autumn season. The only exception here is R for autumn, which slightly decreases for the DYNAMIC approach.

Mean statistics, based on daily values from 6 stations (Harwell, Jarczew, Risø, Tange, Ulborg and Anholt) for 2012, indicate generally better performance (higher FAC2 and R and lower bias) for the DYNAMIC simulation than for BASE (Table 3). FAC2 and R increase moving from BASE to DYNAMIC for winter, spring and summer and bias and error measures are improved for winter and spring (Table 4). The most significant improvement is for the winter season, whereas the smallest changes are for summer. The DYNAMIC simulation has a larger positive bias than the BASE simulation during spring.

A general pattern of the differences between BASE and DYNAMIC, described above, is present at all the individual sites (Fig. 8). The BASE simulation significantly overestimates the measured ammonia concentration for winter, which is improved for the DYNAMIC simulation (RMSE decreases for all stations, Fig. 8). An improvement in correlation coefficient between BASE and DYNAMIC for Harwell, Jarczew and Tange is accompanied by an increase of the RMSE. The highest decrease in the model performance between BASE and DYNAMIC is for autumn for Danish sites.

4 Discussion and conclusion

We have observed an improvement for the DYNAMIC approach for the winter and autumn seasons but for the entire year the modelled hourly ammonia peaks are shifted toward the afternoon hours if compared with measurements. This occurs both for the

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BASE and DYNAMIC simulation, despite a strong peak of emission in mid-day for the DYNAMIC approach. Analysis of the vertical distribution of modelled ammonia concentrations indicates that the main source of ammonia in the air is from the surface, as the highest concentrations are close to the ground and decrease linearly with height.

Moreover, the maximum aerosol concentration appears after the peak of NH₃ concentration, which infers that the ammonia peaks are not related to release of ammonia from aerosols. A similar modelled ammonia concentration pattern to that described above was reported by Wen et al. (2014) for STILT-Chem simulations over southern Ontario in the US, with the limitation that the results were not evaluated against observations. Studies presented by Pinder et al. (2006) for the US emphasise that diurnal profile of ammonia emission is especially important for accurately predicting the concentrations at night, since the decreased atmospheric mixing makes the night especially sensitive to emission changes. Without diurnal variation in the emission, the model overestimates the concentrations at night, which leads to a bias in the average ammonia and ammonium concentration. Pinder et al. (2006) compared the model against mean daily observations and found that in January and July, the seasonally varied inventories significantly improve the predictions of NH₃ and NH₄⁺ concentrations. Our study confirms this improvement for the winter season, for which there is an increase e.g. in correlation coefficient or decrease of bias for all four stations representing different geographical areas and climatological conditions. In our investigation however we go a step further by analysing the hourly and vertical ammonia pattern, and found that there is still an unresolved problem with a discrepancy in time of peaks of the ammonia concentration between modelled and measured concentrations.

We have observed that for each analysed episode (March, April, July and October), the surface ammonia concentrations are anti-correlated with the planetary boundary layer height – midday peaks of PBLH are accompanied by a local minimum of NH₃ concentrations. It has been previously shown that PBLH is an important variable for air quality modelling, which is often difficult to simulate accurately in numerical models (Dabberdt et al., 2004; Hu et al., 2010b; Xie et al., 2012). Determining the PBLH is

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important in atmospheric numerical models, because it is used in other physical param-
eterisations and because it is a governing parameter for the distribution of trace gases
(Geels et al., 2007). Meteorological conditions are known to exert a direct impact on the
air quality simulation. Han et al. (2008) showed that the difference in modelled vertical
turbulent mixing is one of the main reasons for the discrepancy in pollutant concentra-
tion among the chemical transport models. The purpose of a PBL parametrization is to
redistribute energy and humidity in the PBL. Both humidity and temperature have an
impact on ammonia concentrations. Previous studies suggest that the YSU scheme in
WRF (Hong et al., 2006), which was also applied in our study, tends to overestimate
the PBL height (Hu et al., 2010a; Xie et al., 2012). The highest variability between mea-
surements and modelled data are in the midday and are rather constant at night (Ács
et al., 2014). Kim et al. (2013) have found that the YSU and MYJ schemes in WRF
overestimate, while the ACM2 and MYNN underestimate the PBL height. Their study
over Greater Paris indicates that the modelled mean PBLH differs significantly among
the schemes and by more than 300 % between the MYNN and YSU. Overestimation
of PBLH in chemical transport models, like e.g. WRF-Chem might cause an overes-
timation of mixing layer depth and result in an underestimation of modelled pollution
concentrations. The latter might be especially relevant for emissions that are released
from the surface. Based on the meteograms plotted here, we suggest that the verti-
cal extent of the PBL will directly impact the overall concentration of ammonia in the
PBL layer. The PBL physics – and therefore also the choice of parametrisation – must
therefore affect the ammonia concentrations both at the surface as well as throughout
the PBL layer.

The potential higher bias in T2 from the WRF model, reported for Central Europe for
summer season by e.g. Skjøth et al. (2015) and Kryza et al. (2015) will impact the mod-
elled ammonia concentrations. The Jarczew station, located in this region reveals an
increased bias in ammonia concentration for the DYNAMIC simulation in comparison
to the BASE simulation during the summer season. However, this increased bias for
the DYNAMIC simulation is also present for the spring season. This suggests that the

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bias in ammonia concentrations is not related to a bias in temperatures that is mainly seen during summer. Most likely the reason is a combination of a too coarse grid in WRF-Chem and how the ammonia emission is distributed between different emission sectors implemented into the dynamic emission model, which are then affected by the meteorological factors. Several studies showed that significant differences may occur between measured and modelled ammonia concentrations due to the grid size resolution. Especially, the highly localized nature of NH₃ emissions is causing this difference (Dore et al., 2007; Van Pul et al., 2009).

Overall, the results suggest the disagreements between the model and the observations in the hourly profiles are mainly related to emission of ammonia and dilution processes (e.g. increased PBL height and chemical conversion). The results with the aerosols suggest that the high night-time concentrations are not due to release of ammonia from nitrogen containing aerosols. The most likely cause of these high concentrations is that the flux of ammonia from the surface is too high during the night time. The flux of ammonia from agricultural sources away from the surface is mainly dependent on two processes – direct emission due to volatilization (e.g. higher temperatures give higher emission) and the effect of turbulence. The parametrisation we have used does not provide an increase of 100 % in emission from manure that is applied to the field as it is directly linked to previous studies with the ALFAM model (Gyldenkærne et al., 2005; Skjøth et al., 2004) and the parameterisation does not take into account turbulence at the surface but instead uses wind speed data from the ALFAM model (Skjøth et al., 2004). The combination of the temperature and wind speed effect has the consequence that there will be a continuous release of ammonia during night-time even during low temperatures and low wind speeds. Therefore, it must be expected that this limitation causes a redistribution of emission from day to night-time thereby reducing the diurnal emission profile causing a lower daytime peak. If this hypothesis is correct, then new field experiments as a replacement of older experimental data from the ALFAM would be appropriate. Such observations could be used to update the ef-

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sion. In April, for the same areas, much higher concentrations are modelled with the DYNAMIC simulation than with the BASE simulation. This is related to high ammonia emissions during that period caused by application of manure and fertilizers in growing crops. Simultaneously for April, the DYNAMIC simulation calculated the highest ammonia concentrations for many European regions, presented in this study as the number of day with the highest hourly concentrations. Similar findings on spatial variability were also documented by Hamaoui-Laguel et al. (2014), for a study conducted over France. Their study focused on the effect of emission from mineral fertilizers and these results showed that the spatial pattern was highly dependent on actual meteorological conditions. These results using the CHIMERE model, the results by Wen et al. (2014) with STILT-Chem, our results with WRF-Chem as well as previous studies with this emission model all suggest that it is important to have a direct connection between hourly meteorological variables and the level of ammonia emission.

Our study does not include bi-directional exchange, which can further influence the modelled ammonia concentrations. Recent study provided by Zhu et al. (2015) suggests that although the implementation of bi-directional exchange leads to a better fundamental description of NH₃ emissions from fertilizers, it does not uniformly improve estimation of NH₃ concentrations, NH₄⁺ wet deposition and nitrate aerosol concentrations. However, Bash et al. (2013) reported that an implementation of bi-directional exchange of NH₃ improved the simulations of NH_x wet deposition and improved the simulation of ambient nitrate aerosol concentrations for the US. Wichink Kruit et al. (2012) showed that with the new description in the LOTOS-EUROS model, which includes bi-directional surface–atmosphere exchange, the modelled ammonia concentrations increase almost everywhere, in particular in agricultural source areas. The reason for this is that by using a compensation point the ammonia lifetime and transport distance is increased. A comparison with measurements shows that the model results better represent measured ammonia concentrations; however the concentrations in nature areas are slightly overestimated, while the concentrations in agricultural sources are underestimated.

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Our study indicates that the current description of the diurnal cycle of the ammonia concentration from fields is not sufficiently accurate and more research is needed in order to improve the processes that describe the emission from fields. The results suggest that the governing processes in relation to the diurnal cycle are the atmospheric mixing and the stronger daily pattern of ammonia emission, e.g. through increased evaporation or increased fluxes from the surface. The latter is still quite difficult to observe as only very few sites measure ammonia concentrations with a sufficiently high temporal resolution to that required for a proper evaluation and developments of better emissions models. New field experiments as a replacement of older experimental data would be appropriate in order to update an observational assessment of the emission process parameterised in the model. This is one of the objectives of the ECLAIRE project.

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Table 1. Model components and configuration.

Category	Model setup
Simulation period	Jan–Dec 2012
Domains	Europe, 161 × 131 grids
Horizontal resolutions	36 km × 36 km
Vertical resolution	48 layers
Shortwave and Longwave radiation	RRTMG
Land-surface model	Noah LSM
Boundary layer scheme	YSU
Cumulus parameterization	Grell and Denvenyi (2002)
Microphysics	Lin et al. (1983)
Chemistry	RADM2 and MADE/SORGAM with aqueous reactions

Please refer to the WRF and the WRF-Chem user's guides for a complete description of the options.

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Table 2. Mean statistics based on hourly data (N) for Harwell for 2012: FAC2 (factor of two), MB (mean bias), NMB (normalised mean bias), RMSE (root mean squared error), R (correlation coefficient).

Season	N	FAC2	BASE			RMSE $\mu\text{g m}^{-3}$	R
			MB $\mu\text{g m}^{-3}$	NMB $\mu\text{g m}^{-3}$	NMGE $\mu\text{g m}^{-3}$		
winter	1491	0.60	0.44	0.60	0.98	1.31	0.36
spring	1949	0.47	0.91	0.35	0.82	2.94	0.44
summer	1156	0.36	0.35	0.33	0.97	1.33	0.22
autumn	1523	0.50	0.56	0.69	1.02	1.15	0.26
DYNAMIC							
winter	1491	0.64	-0.12	-0.16	0.52	0.63	0.45
spring	1949	0.47	2.26	0.87	1.23	5.60	0.57
summer	1156	0.35	0.52	0.49	1.07	1.50	0.28
autumn	1523	0.51	0.46	0.56	0.94	1.06	0.24

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Table 3. Mean statistics for NH₃ concentrations based on daily in (*N*) observations from 6 sites (Harwell, Jarczew, Risø, Tange, Ulborg and Anholt) for 2012.

	<i>N</i>	FAC2	MB μg m ⁻³	NMB μg m ⁻³	NMGE μg m ⁻³	RMSE μg m ⁻³	<i>R</i>
BASE	2020	0.38	0.93	1.10	1.28	1.76	0.55
DYNAMIC	2020	0.42	0.85	1.00	1.23	1.94	0.66

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Table 4. Mean statistics split into seasons for NH₃ concentrations based on daily observations (*N*) from 6 sites (Harwell, Jarczew, Risø, Tange, Ulborg and Anholt) for 2012.

Season	<i>N</i>	FAC2	MB μg m ⁻³	BASE			RMSE μg m ⁻³	<i>R</i>
				MGE μg m ⁻³	NMB μg m ⁻³	NMGE μg m ⁻³		
winter	502	0.22	1.18	1.21	3.65	3.75	2.10	0.34
spring	539	0.46	1.34	1.60	0.87	1.04	2.34	0.57
summer	491	0.46	0.57	0.72	0.69	0.86	0.98	0.60
autumn	488	0.37	0.58	0.76	0.92	1.19	1.09	0.46
DYNAMIC								
winter	502	0.38	0.20	0.34	0.61	1.06	0.49	0.42
spring	539	0.46	1.71	1.97	1.11	1.28	3.22	0.66
summer	491	0.49	0.56	0.71	0.67	0.85	0.97	0.63
autumn	488	0.33	0.84	1.06	1.32	1.65	1.72	0.27

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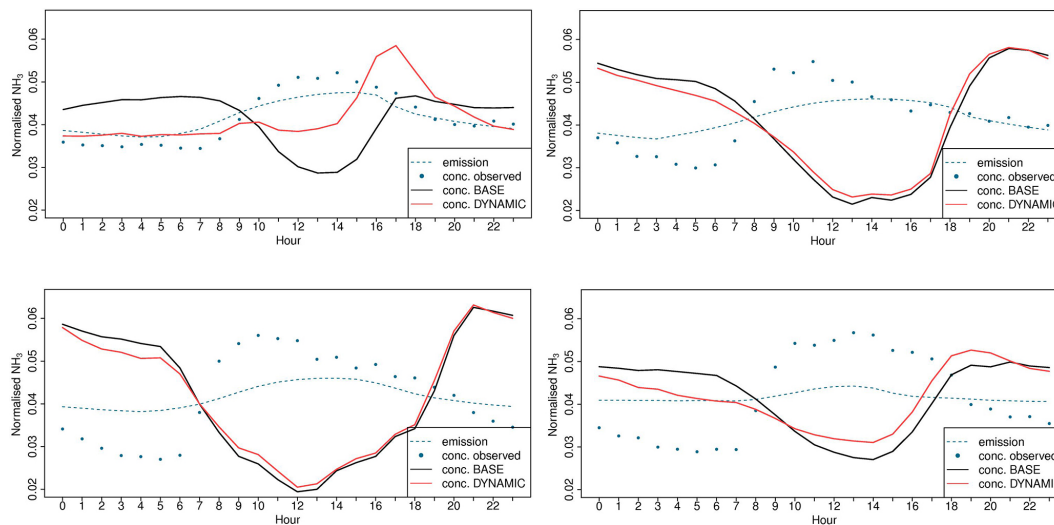


Figure 1. Normalised NH₃ concentrations and emission according to hours for four seasons for the grid corresponds to the Harwell station. Normalisation procedure: the sum of emission/concentration for each individual hour (0–23) was divided by the total sum of emission/concentration in the season.

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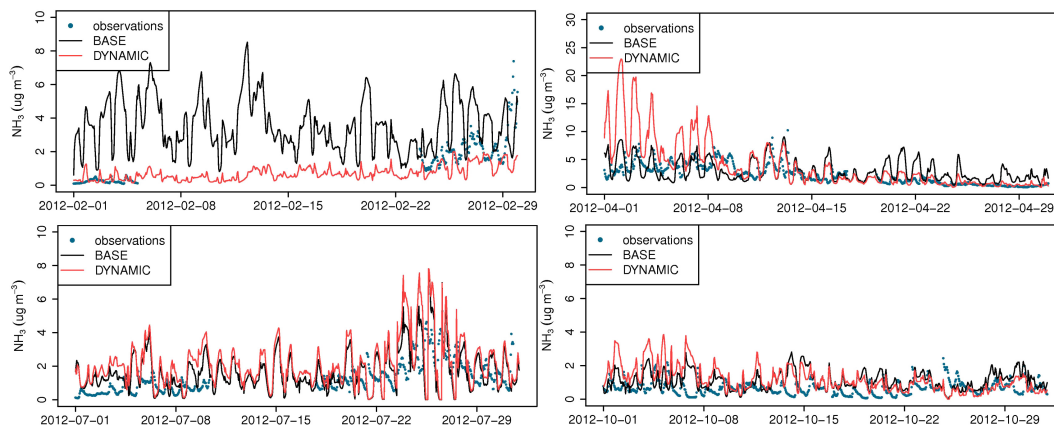


Figure 2. Modelled and observed hourly NH₃ concentrations for Harwell for February, April, July and October 2012. Please notice there is a different y scale for April.

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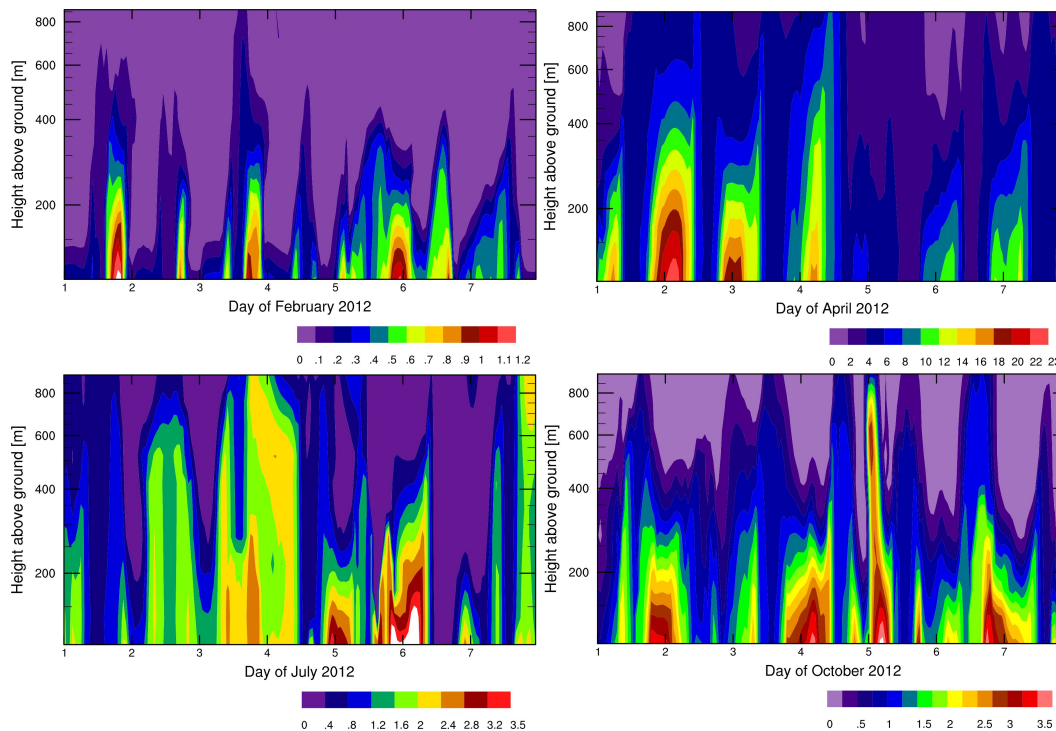


Figure 3. Temporal and vertical distribution of NH₃ concentrations [$\mu\text{g m}^{-3}$] for the DYNAMIC scenario.

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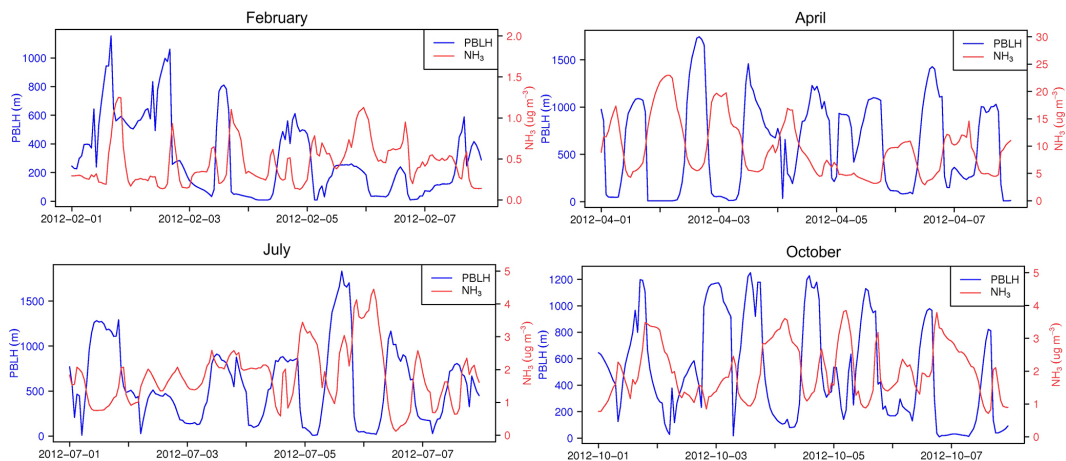


Figure 4. Time series of modelled (DYNAMIC) PBLH and NH₃ concentrations for the episode of February, April, July and October 2012.

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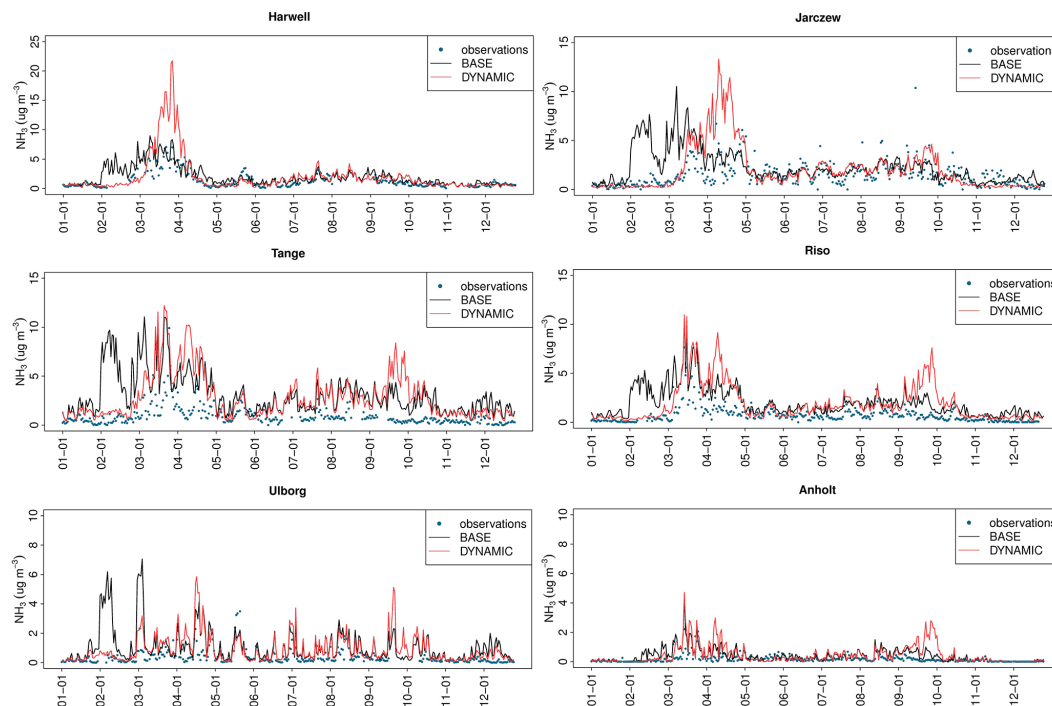


Figure 5. Modelled and observed daily NH₃ concentrations for 2012. Different y axis scales are used for the different stations.

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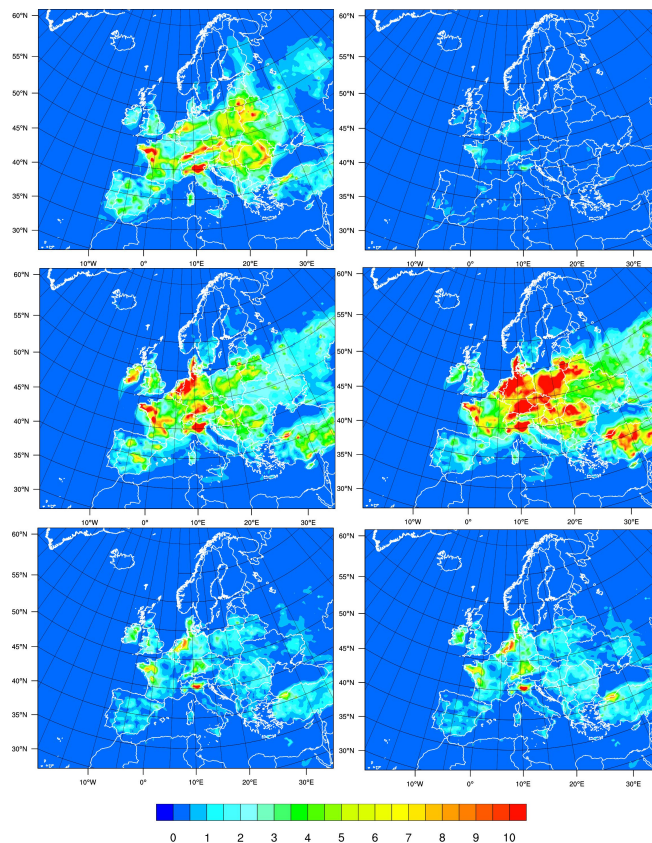


Figure 6. Spatial distribution of NH₃ concentrations on 15 February (upper), 15 April (middle), and 15 October (lower) 2012 at 12.00 a.m. BASE scenation – left column, DYNAMIC scenario – right column. Units: $\mu\text{g m}^{-3}$.

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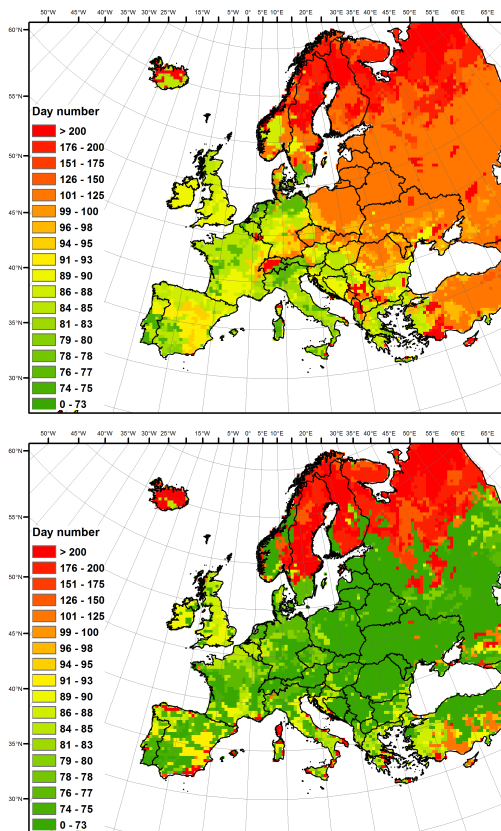


Figure 7. Julian day number in the year 2012, for which the model calculated the highest hourly ammonia concentrations – upper figure for DYNAMIC and lower for BASE.

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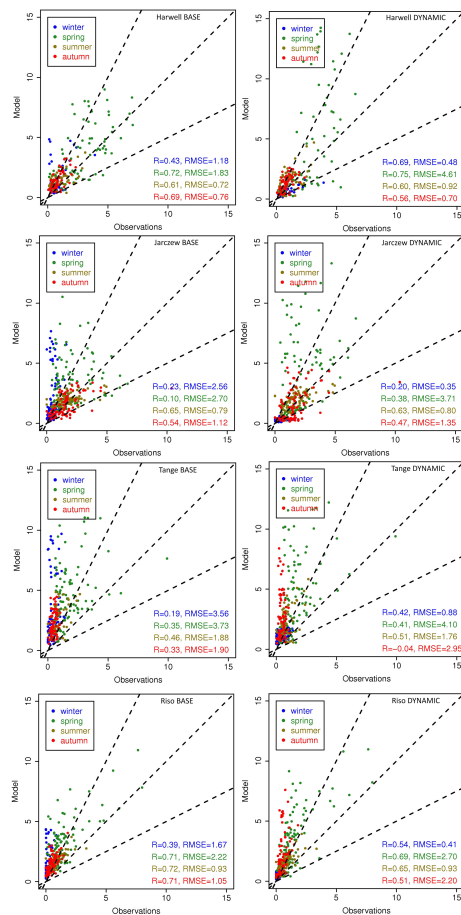


Figure 8. Scatter plots of NH₃ concentrations based on daily values for individual sites for the BASE (left) and DYNAMIC (right) simulations and according to seasons.

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